

GOVT. POLYTECHNIC, JAGATSINGHPUR

CIVIL ENGINEERING DEPARTMENT

**LEARNING MATERIAL OF BUILDING MATERIAL &
CONSTRUCTION TECHNOLOGY**

3RD SEMESTER

FACULTY NAME – ANANTA BISWAL

Building Material & construction Technology



1. Stone
2. Brick
3. Cement
4. Mortar
5. Aggregate
6. concrete
7. OCM
8. Surface
Protective
materials.

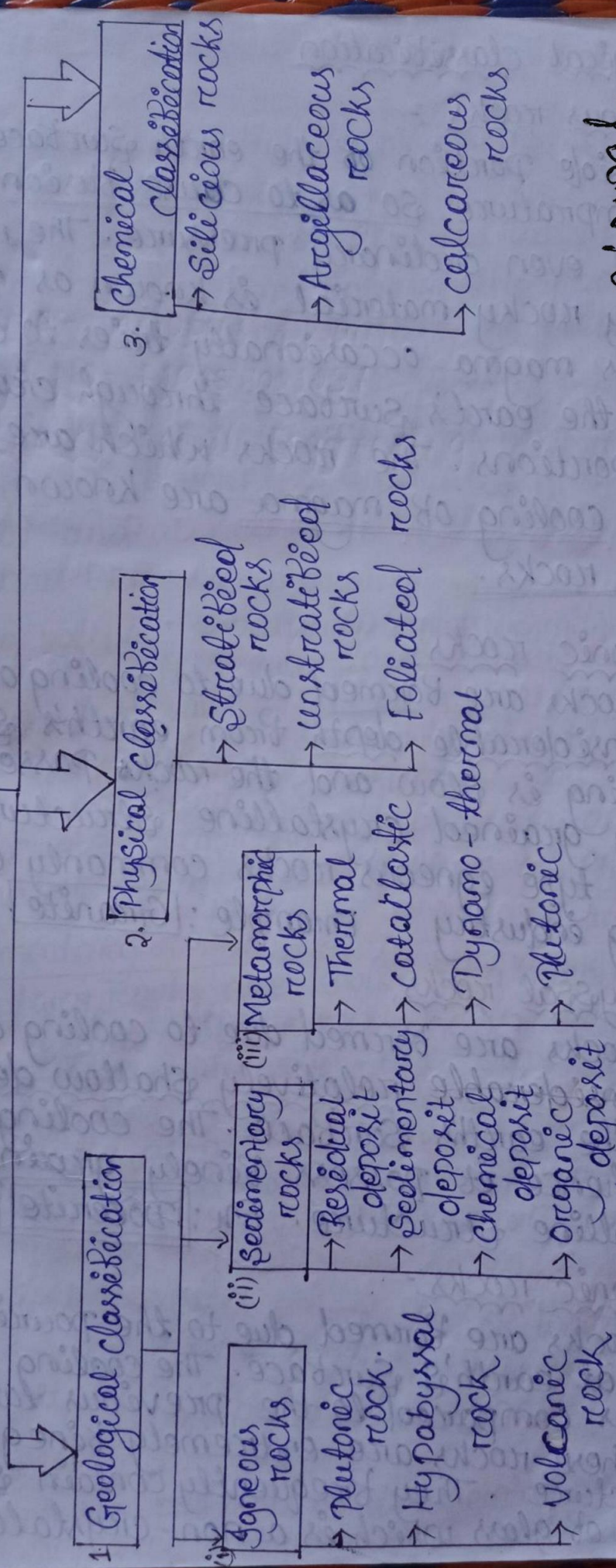


1. Foundation
2. wall
3. Masonary works
4. Door
5. Window
6. Lintels
7. Floor
8. Roof
9. Stairs
10. DPC
11. Plastering
12. Pointing
13. Painting
14. white washing

Ananta Biswal

The building stones are obtained from rocks which are classified in 3 ways.

Classification of Rocks



Ananta Biswal

① Geological classification

(i) Igneous rocks :-

The inside portion of the earth's surface has high temperature, so as to cause fusion by heat at even ordinary pressure. The molten or partly rocky material is known as magma and this magma occasionally tries to come out to the earth's surface through cracks (or) weak portions. The rocks which are formed by the cooling of magma are known as the igneous rocks.

(a) Plutonic rocks :-

Such rocks are formed due to cooling of magma at a considerable depth from earth's surface. The cooling is slow and the rock possess coarsely grained crystalline structure. Plutonic type igneous rocks commonly used in building industry. Example: Granite

(b) Hypabyssal rocks :-

Such rocks are formed due to cooling of magma at a relatively shallow depth from the earth's surface. The cooling is quick hence it possess finely grained crystalline structure. Ex: Dolerite

(c) Volcanic rocks :-

Such rocks are formed due to the pouring of magma at earth's surface. The cooling is very rapid as compared to the previous two cases. Hence these rocks are extremely fine grained in structure. They frequently contain some quantity of glass which is a non-crystalline

material. En: Basalt

Note:

★ The principal constituents of magma are quartz, mica and Feldspar.

★ When magma cools rapidly, its mass expands under the pressure of intensively evolving gases. Subsequent rapid cooling of swollen lumps of magma gives rise to glassy porous rock known as pumice used as aggregate for light wt. concrete, as heat insulating material and as active mineral admixture to lime and cements.

★ During volcanic eruption ashes, sands are mixed with molten lava to form tuff lava. Comented tuff lava is called volcanic tuff. Tuffs have a glassy structure due to rapid cooling & are used as aggregate for light wt. concrete & mortar and an active admixture to air-setting lime or cement.

(ii) Sedimentary rocks:

Sedimentary rocks are known as a queous or stratified rocks. The various weathering agencies, e.g. rain, sun, air, frost etc. break up carried away from their place of origin by the agents of transport such as frost, rain, wind, flowing water etc.

Following 4 types of deposits occur.

(a) Residual deposits:

Some portion of the products, of weathering remain at the site of origin, called as residual deposits. (Sedimentary deposit)

(b) Sedimentary deposits:

The insoluble products of weathering are carried away in suspension and when such products are

deposited they give rise to the sedimentary deposits.

(c) Chemical deposits:
Some materials that is carried away in solution may be deposited by some physio-chemical process such as evaporation, precipitation etc. It give rise to chemical deposits. \rightarrow hydrolysis, carbonation.

(d) Organic deposits:
Some portion of the product of weathering gets deposited through the agency of organisms such deposits are known as organic deposits.
Example of sedimentary rocks = Gravel, Sandstone, lime stone, gypsum, lignite etc.

(iii) Metamorphic Rocks

\rightarrow These rocks are formed by the change in character of the pre-existing rocks. The igneous as well as sedimentary rocks are changed in character when they are subject to great heat and pressure. The process of change is known as metamorphism.

\rightarrow There are 3 agents of metamorphism, namely, heat, pressure and chemically acting fluids. The heat may be supplied by the general rise of temperature with depth or by igneous magma. The pressure may be developed due to load of rocks or movement of earth. The chemically acting fluids play a passive role only & they don't take active part in the process of metamorphism.

4 type of metamorphism occur with various combinations of heat & pressure

uniform direction

a Thermal metamorphism:-

Heat is predominant factor in this type of metamorphism.

b cataclastic metamorphism:-

At the surface of the earth, the temperature are low and metamorphism is brought about by directed pressure only. Such metamorphism is known as cataclastic metamorphism.

c Dynamo-thermal metamorphism:-

There is a rise in temperature with increase in depth. Hence the heat in combination with stress brings about the changes in rock. Such metamorphism is called Dynamo-thermal metamorphism.

d Plutonic metamorphism:-

Metamorphic changes at great depths are therefore brought about by uniform pressure and heat. Such metamorphism is known as the plutonic metamorphism.

2. Physical Classification:-

(i) Stratified rocks:-

Show distinct layers along which the rocks can be split. Ex: sand stone, limestone, slate, shale, marble etc. [i.e. Sedimentary rocks]

(ii) Unstratified rocks:-

do not show any stratification and can't be easily split into thin layers. Ex: Granite, basalt, trap etc.

(iii) Foliated rocks:-
have a tendency to split up only in a definite direction. Most of the metamorphic rocks have a foliated structure except for quartzite, marble which have granulose structure.

3 Chemical classification:-

(i) Siliceous rocks:-

Principal constituent is silica (SiO_2) i.e. sand.

These rocks are very hard and durable.

Ex - Granite, basalt, trap, quartzite, gneiss, Syenite etc.

(ii) Argillaceous rocks:-

Principal constituent is clay or argil (Al_2O_3).

These rocks are hard & brittle.

Ex: slate,

(iii) Calcareous rocks:-

In these rocks lime is predominates.

Ex: limestone (CaCO_3), marble

Use of stones:-

- stones are extensively used for the construction of foundations, walls, columns and arches in buildings.
- They are ideally suited for the construction of retaining walls, bents, piers of bridges and dams.
- Polished granite and marble are used for the face works of important buildings.
- Stone slabs are used for flooring, damp-proof course, lintels, roofing and pavers round the buildings, as well as for footpaths.

- Crushed stones are used as a basic inert material in concrete, for making artificial building blocks, such as railway ballast and to provide base course for roads.

Natural bed stone:-

→ The building stones which are obtained from rocks, have a distinct plane of division along which stones can easily be split. This plane is known as the natural bed of stone.

Importance: In stone masonry, the general rule to be observed is that the direction of natural bed of all sedimentary stones should be perpendicular (or) nearly so to the direction of pressure. Such an arrangement gives maximum strength to the stonework.

* The natural bed of stones can be detected by pouring water and examining the direction of layers. The magnifying glass may also be used for this purpose. The stone breaks easily along these natural beds.

→ The natural beds should be placed in horizontal in walls construction.

→ In stone arches, the stones are placed with their natural beds radial.

Qualities of good building stone:-

→ Crushing strength \geq 100 MPa.

→ uniform colour, free from clay holes, spots of other colour, bands etc.

→ Stones should be durable & for making stones

durable, their natural bed should be carefully noted. The stone should be arranged in a structure that the natural bed is \perp or nearly so to the dirⁿ of pressure.

* \rightarrow For road work, co-eff of hardness > 17
co-eff of hardness $14-17 \Rightarrow$ medium hardness
co-eff of hardness $< 14 \Rightarrow$ Low hardness

* \rightarrow In attrition test, $\text{wear} > 3\% \Rightarrow$ not satisfactory
 $\text{wear} = 3\% \Rightarrow$ Tolerable

(For good building stone wear $\neq 3\%$ (i.e) wear $\leq 3\%$)

\rightarrow About 6 to 12 months is considered to be sufficient for proper seasoning to remove some moisture or quarry sap.

(\rightarrow Specific gravity > 2.7

\rightarrow Toughness index $< 13 \Rightarrow$ Not tough

Toughness index $13-19 \Rightarrow$ Moderately tough

Toughness index $> 19 \Rightarrow$ Highly tough.

* % absorption by weight after 24 hrs $\neq 0.6$.

Attrition \rightarrow It is the friction betⁿ same type of material.

Abrasion \rightarrow Friction betⁿ different type of material.

1 gallon = 3.78 lit. Bricks

1 foot = 3.17 mm

A human brain has a capacity to store 5 times as much information as Wikipedia

1 pound = 0.4536 kg.

Standard size of bricks = (19 x 9 x 9) cm

Modular size of bricks = (20 x 10 x 10) cm \Rightarrow For High class brick masonry.

Size of Frog = (10 x 4 x 1) cm \Rightarrow To form key for holding the mortar.
 \hookrightarrow 1-2 cm deep
and therefore the bricks are laid with frogs on the top.

non-standardize (or) traditional (or) field bricks = (23 x 11.4 x 7.6) cm

Bricks are of 2 types \Rightarrow ① Burnt bricks
② Unburnt bricks (or) sundried bricks

Burnt bricks are used for construction work & classified into 4 main categories.

1st class brick

\rightarrow Table moulded & burnt in kilns.

\rightarrow Deep red, cherry (or) copper colour.

\rightarrow Free from flaws, cracks and stones & smooth rectangular surfaces with straight edges. Uniform texture & no impression when scratch by bingert nail.

2nd class brick

\rightarrow Ground moulded and burnt in kilns.

\rightarrow Deep red, cherry (or) copper colour.

\rightarrow Small cracks & distortion are permitted. Uniform texture and no impression when scratch by bingert nail.

3rd class brick

\rightarrow Ground moulded and burnt in clamps.

\rightarrow Soft and reddish yellow coloured.

\rightarrow rough surfaces with irregular & distorted edges.

→ Metallic & ringing sound when struck against each other

→ Metallic and ringing sound when struck against each other

→ produce dull sound when struck with each other

★ Water absorption = 12-15% of its dry weight

★ Water absorption = 16-20% of its dry weight

★ W.A = 25% of its dry wt.

★ crushing strength $\geq 10 \text{ MPa}$ or N/mm^2

★ crushing strength $\geq 7 \text{ MPa}$

Use \Rightarrow pointing, exposed face work in masonry, flooring.

Use \Rightarrow For all important or unimportant heavy masonry work & R.C.C. structure.

Use \Rightarrow For important and temporary structure and at place where rainfall is not heavy.

R.B. work
Weight = 3.85 K.g.

R.B. work \rightarrow Rain Bounce Brick work

★ Percentage of water absorption = $\frac{dw}{Sw} \times 100$

4th class Brick \Rightarrow over burnt and badly destroyed in Thama bricks

Shape & Size brittle in nature.

Use \Rightarrow Ballast of such brick is used for foundation & floors in lime concrete and road metal, because of fact that over burnt bricks have a compact structure and hence they are sometime found to be stronger than even 1st class bricks.

Properties of good brick :

A good brick should have \rightarrow

1. uniform size, shape, rectangular surfaces with parallel sides and sharp straight edges.

2. uniform deep red and cherry colour as indicative of uniformity in chemical composition and thoroughness in the burning of brick.

3. Surface shouldn't be too smooth to cause slipping of mortar. The brick should have precompact and uniform texture. A fractured surface should not show fissures, holes, grits (or) lumps of lime. \Rightarrow Texture & compactness.
4. Brick should be so hard that when scratched by a fingernail no impression is made. \Rightarrow Hardness
5. Water absorption $\leq 20\%$ of its dry wt. when kept immersed in water for 24 hrs.
6. Crushing strength > 10 MPa.
7. Soundness \Rightarrow Metallic sound be produced when 2 bricks struck together
8. Brick earth should be free from stones, grits, organic matter etc.

Composition of good brick earth:

Silica $\Rightarrow 50-60\%$
 Alumina $\Rightarrow 20-30\%$
 Lime $\Rightarrow 10\%$

In cement
 Lime (CaO) $\Rightarrow 60-65\%$
 Silica $\Rightarrow 17-20\%$
 Alumina $\Rightarrow 3-8\%$

Alkalies $\Rightarrow < 10\%$

Ferrous oxide $\Rightarrow < 7\%$ [$< 20\%$]

Magnesia $\Rightarrow < 1\%$

$\text{CO}_2, \text{SO}_3 \text{ \& \ } \text{H}_2\text{O} \Rightarrow \text{Traces}$

LSA \Rightarrow cement

SAL \Rightarrow Brick

Silica:

- \rightarrow Exists in clay either as free or combined. As free sand is mechanically mixed with clay and in combined form, it exists in chemical composition with alumina.
- \rightarrow It prevents cracking, shrinking and warping of raw bricks. Thus it imparts durability and uniform shape to bricks.
- \rightarrow Excess of silica destroys the connection betⁿ particles and the bricks become brittle.
- \rightarrow A large percentage of sand (or) uncombined

Silica in clay is undesirable. However it is added to decrease shrinkage in burning and to increase the refractoriness of low alumina clays.

2. Alumina [20-30%]

* Alumina is chief constituent, of every kind of clay.

→ It absorbs water and imparts plasticity to the earth such that it can be moulded.

→ If alumina is present in excess, it causes cracks in bricks on drying and becomes too hard when burnt.

→ Clays having high alumina are bound to be very refractory.

Lime [$< 10\%$] [but $< 5\%$ most desirable for good brick earth]

→ Prevents the shrinkage on drying.

→ It causes silica in clay to melt on burning and hence helps to bind it. But the excess lime causes the brick to melt and hence its shape is lost.

* The lumps of lime are converted into quicklime after burning and this quicklime slakes and expands in presence of moisture; such an action results in splitting of bricks into pieces.

→ In carbonated form, lime lowers the fusion point.

Oxides of iron [Ferric oxide] [$< 7\%$]

→ It gives red colour on burning when excess of oxygen is available and dark brown (or) even black colour when oxygen available is insufficient, but excess of ferric oxide makes the brick dark blue (or) blackish.

- It improves impermeability and durability.
- It lowers the fusion point of the clay, especially if present as ferrous oxide.
- It gives strength and hardness.

Magnesia: [$< 1\%$]

- decrease shrinkage and gives yellow tint. → Excess of magnesia leads to the decay of bricks.

* Harmful ingredients in Brick earth:-

1. Lime:-

- Excess of lime → colour of brick changes from Red to yellow.
- When lime is present in lumps, it absorbs moisture, swells, cause pieces of brick.

2. Iron pyrites:-

- Iron pyrites are present in brick earth, then bricks tend to oxidise and decompose during burning and may split into pieces.
- Pyrites discolourise the brick.

3. Pebbles, gravels, grits:- It doesn't allow clay to be mixed uniformly and thoroughly, which will result in weak and porous bricks.

- Bricks containing pebbles will not break regularly as desired.

4. Alkalies:-

- Excess of alkalies causes brick to melt and lose their shape.

- Alkalies causes efflorescence - when brick come in contact with moisture.

Water is absorbed and the alkalies crystallise. On drying, the moisture evaporates, leaving behind grey (or) white powder deposits on the brick which spoil the appearance and this phenomenon is called efflorescence.

5. Organic matter:

→ During burning of bricks, organic matter gets burnt completely, leaving behind pores and hence making bricks porous.

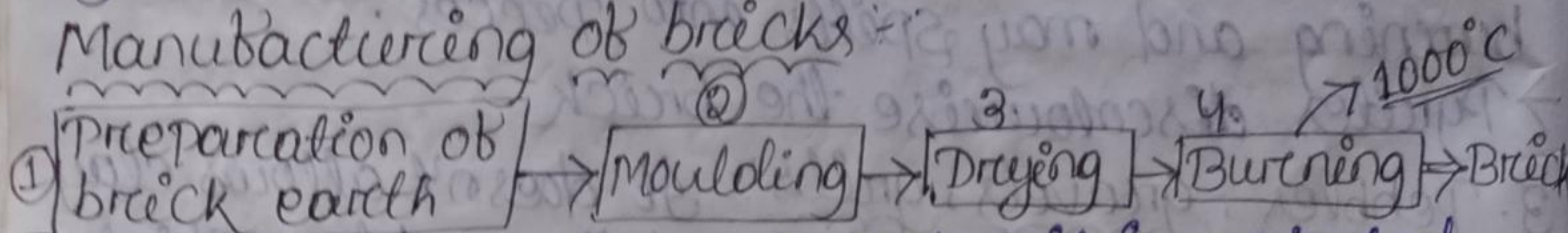
→ It also absorbs water and therefore reduces strength of bricks.

6. Water: Large amounts of free water causes shrinkage of bricks on drying, whereas combined water causes shrinkage during burning.

7. Sulphur:

It sulphur is present in brick earth & insufficient time is given (during burning) for oxidation of carbon & sulphur, then sulphur will cause the formation of spongy, swollen structure in the brick and the brick will be discoloured by white blotches.

Manufacturing of bricks:



- Unsoiling → Top 20 cm depth of soil is rejected because of full impurities.
- Digging → Clay is then dug out & make heap of soil of height 60 cm to 120 cm.
- Cleaning → If stones, pebbles are excess, then clay is to be washed & screened.
- Weathering → Clay is then exposed to atmosphere for softening (or) mellowing from few weeks to full season of monsoon.
- Blending → Proper blending done by clay is made loose & turning up & down in vertical dirⁿ.
- Tempering → Water in req^d quantity is added to clay and the whole mass is kneaded (or) pressed under the feet of men or cattle.

* For manufacturing of good brick, tempering is done in pug mills and operation is called pugging.

* In pug mill, feeding of clay from top and taking out of pugged clay from bottom are done simultaneously.

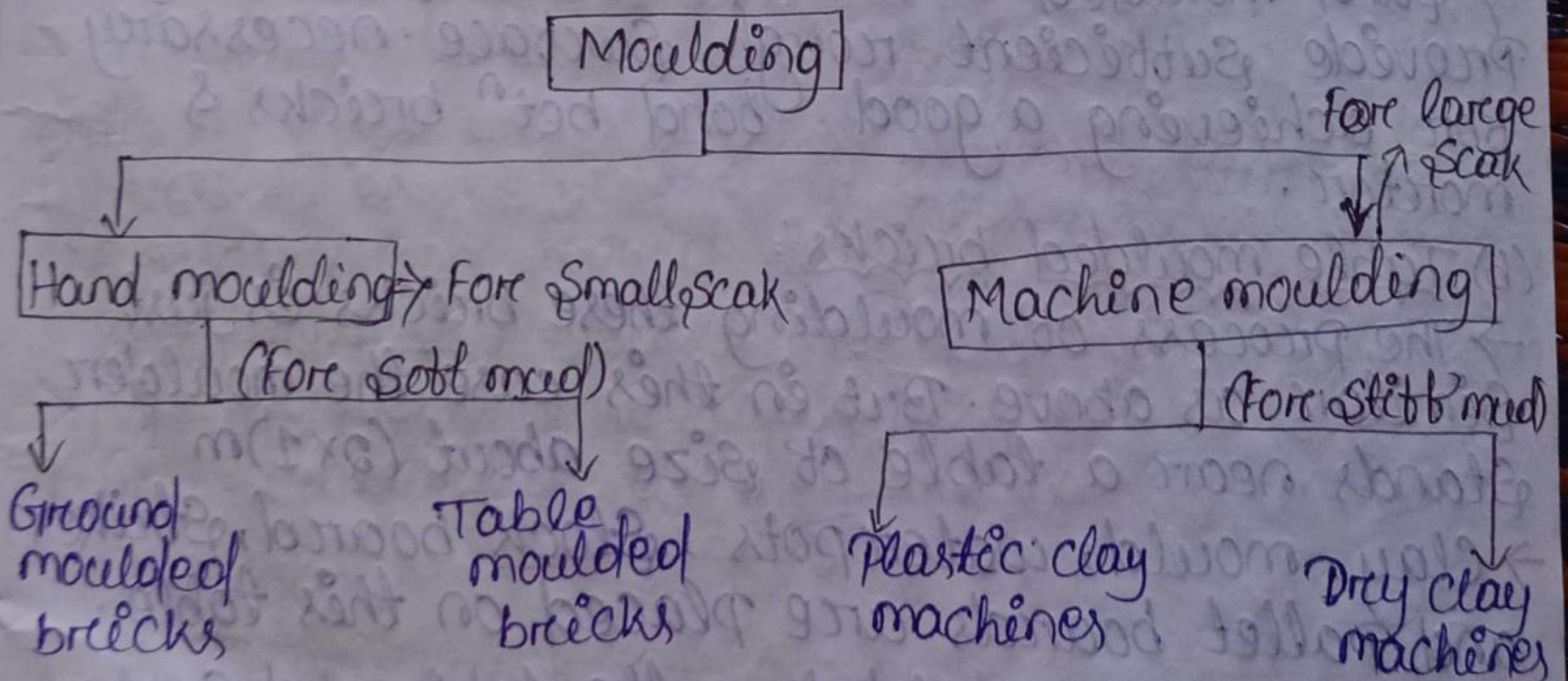
⊙ Tempering of clay is done such that brick earth can be rolled without breaking in small threads of 3mm diameter.

2. Moulding :-

→ It is the process of giving required shape to the brick from prepared brick earth.

* We know that bricks shrink during drying and burning. therefore, moulds are made 8-12% larger size than of fully burnt bricks.

→ Exact percentage of increase in dimensions of mould is found by experiments on brick earth at site.



(i) Hand moulding :-

→ Adopted, when man power is cheap, & easily available for manufacturing of bricks, usually done on small scale.

→ Moulds are rectangular boxes of wood or steel, open at top & bottom.

(a) Ground moulded bricks:

- adopted when a large & level land is available (ii)
- In this process ground is levelled & sand is sprinkled over it.
- Extra (or) surplus clay is removed either by a wooden strike (or) metal strike.
- Now mould is lifted and raw bricks are left on the ground.
- To prevent the moulded bricks from sticking to the side of the mould, sand is sprinkled on the inner side of mould, (or) mould may be dipped in water every time before moulding is done.
- ^{v.v.f.} → If sand is sprinkled then bricks are called as sand-moulded bricks, & in case of mould dipped in water, bricks are called as slip-moulded bricks.

→ Sand moulded bricks are better since they provide sufficient rough surface necessary for achieving a good bond betⁿ bricks & mortar.

(b) Table moulded bricks:

- The process of moulding these bricks is just similar as above. But in this case the moulder stands near a table of size about (2x1)m
- Clay, mould, waterpots, stock board, strikes and pallet boards are placed on this table.
- Bricks are moulded on the table and sent for the further drying.
- Cost of brick moulding also increase when table moulding is adopted.

(ii) Machine moulding:

→ Moulding is done with the help of machines when bricks are req^d in a large no. of quantity in a short span of time.

2 ways of classification

(a) Plastic clay machine method:

→ In this method lugged stiff clay is forced through a rectangular opening of brick size by means of an auger.

→ These bricks come out of the opening in the form of a bar and further these bricks are cut into strips by wire lined in frames.

* As the bricks are cut by wire, they are also known as wire-cut bricks.

IES Conv.

(b) Dry clay machine method:

→ In this method, moist, powdered clay is fed into the mould on a mechanically operated press, where it is subjected to high pressure and the clay in the mould takes the shape of bricks.

→ Such pressed bricks are more dense, smooth and uniform than ordinary bricks. These are burnt carefully as they are likely to crack.

3. Drying :-

→ If Green bricks (containing 7-30% moisture) are burnt, are likely to be cracked and distorted.

→ Therefore moulded bricks are dried to remove moisture for controlling of shrinkage and saving of fuel during burning.

→ It is a normal practice to dry moulded bricks to approx m.c of 5-7%.

Drying

Natural

Stage

- ① 1st bricks are allowed to dry in position as released from the mould.
 - ② Then they are turned to stand on edge.
 - ③ Arranged in rows in racks with spaces of less than 1cm betⁿ the bricks.
- Stage ① & ② requires 2 to 3 days and stage ③ about one week.

Artificial

Adopted when bricks are dried on large scale.

Tunnel driers

Hot floor driers

* Tunnel driers are more economical than hot floor driers and they may be periodic (or) continuous.

* Temperature maintained by driers is usually less than 120°C and the process of drying of bricks takes about 1 to 3 days.

* The bricks in stacks should be arranged in such a way that sufficient air space is left betⁿ them for free circulation of air. It is to be seen that bricks are not directly exposed to the wind (or) sun for drying. Suitable screens, if necessary, may be provided to avoid such situation.

4. Burning:-

→ It is very important process in manufacturing of bricks as it provides hardness and strength to the bricks and makes them dense and durable.

* Bricks should be burnt properly. If bricks are overburnt, they will be brittle and hence break easily. If they are underburnt, they will be soft and hence can not carry loads.

Burning of bricks can be divided into 3 stages.

1. Dehydration ($400-650^{\circ}\text{C}$) \Rightarrow Also called water smoking stage.

\rightarrow The water which has been retained in pores of the clay after drying is driven off and the clay loses its plasticity.

- * Some of carbonaceous matter is burnt.
- * A portion of sulphur is distilled from pyrites.
- * Hydrrous minerals like ferric hydroxide are dehydrated.
- * carbonate minerals are more (or) less decarbonated.

2. Oxidation period ($650-900^{\circ}\text{C}$)

\rightarrow Remained of carbon is eliminated.
 \rightarrow The ferrous iron is oxidized to the ferric form.

* Removal of sulphur is completed only after the carbon has been eliminated.

3. Vitrification:

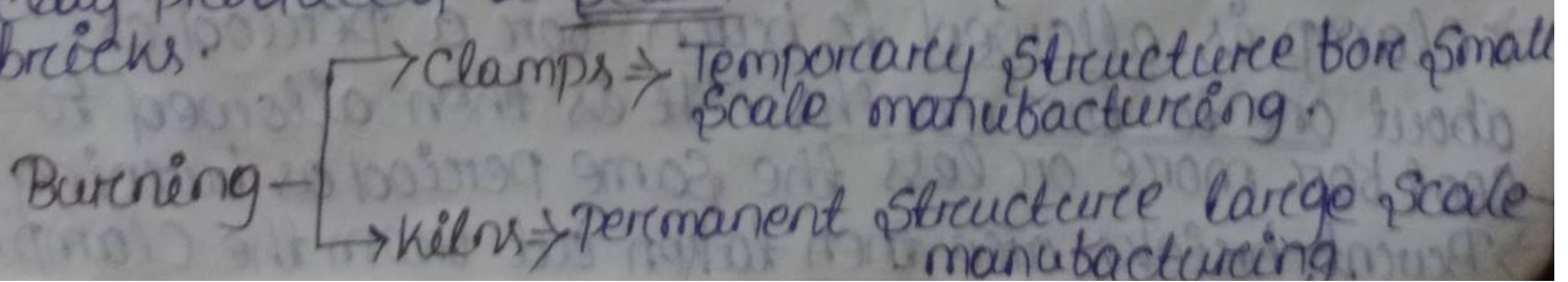
\rightarrow Convert the mass into glass like substance - the temp ranges from $900-1100^{\circ}\text{C}$ for low melting clay and $1000-1250^{\circ}\text{C}$ for high melting clay.

* [1000°C reqd for burning of bricks (avg)]

* A special care is required in cooling the bricks below the cherry red heat in order to avoid checking and cracking.

Note:

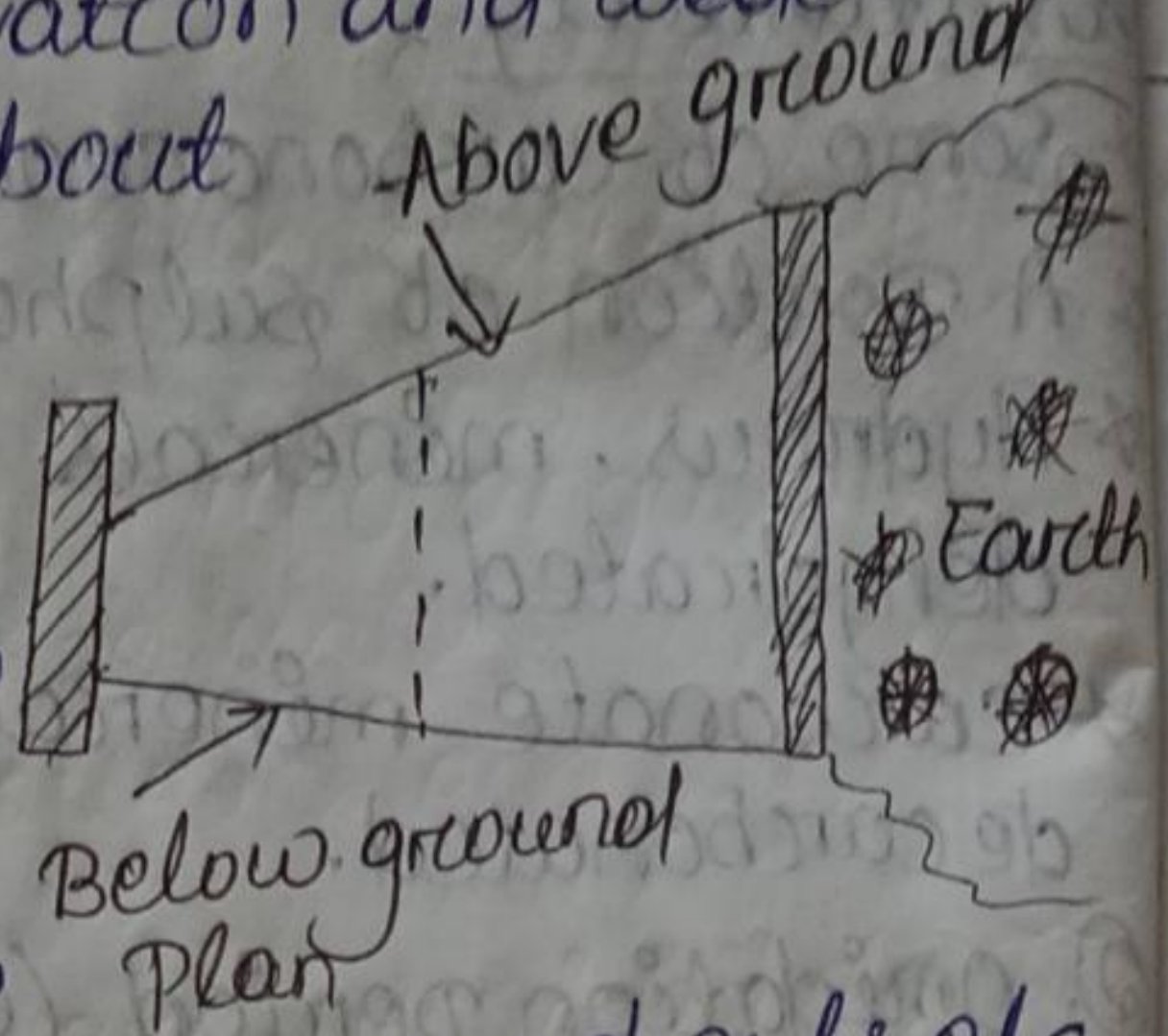
Too rapid heating causes cracking (or) bursting of the bricks. On the other hand, if alkali is contained in the clay (or) sulphur is present in large amount in the clay, too slow heating of clay produced a scum on the surface of the bricks.



(c) Clamp (or) Pazawah:

→ A piece of land is selected such that its shape in plane is generally trapezoidal and floor of clamp is prepared in such a way that short end is slightly in the excavation and wider end is raised at angle of about 15° from ground level.

→ Brick wall in mud is constructed on the short end and a layer of fuel is laid on the prepared floor.



→ Fuel may consist of grass, cow dung, litter, rice husks, ground nuts etc.

→ Thickness of this layer is about 70 cm to 80 cm.

→ Layer consisting of 4 (or) 5 courses of raw bricks, is then put up. The bricks are laid on edges with small spaces betⁿ them for circulⁿ of air.

→ 2nd layer of fuel is then placed and over it, another layer of raw bricks is put up.

→ Alternate layer of fuel and raw bricks are formed.

⊛ Thickness of fuel layer gradually decreases as the height of clamp increases.

→ Total height of clamp is about 3m to 4m. When nearly $\frac{1}{3}$ rd height is reached, the lower portion of the clamp is ignited.

→ When clamp is completely constructed, it is plastered with mud on sides and top and filled with earth to prevent the escape of heat.

→ Clamp is allowed to burn for a period of about one to two months. It is then allowed to cool for more or less the same period as burning.

→ Burnt bricks are then taken out from the clamp.

Advantages of clamp burning:

- Burning & cooling of bricks are gradual in clamps. Hence the bricks produced are tough & strong.
- Burning of bricks by clamps proves to be cheap & economical. No skilled labour and supervision is reqd for the construction & working of clamps.
- Clamp is n't liable to injury from high wind or rain.
- There is considerable saving of fuel.

Disadvantages of clamp burning:

- Bricks are not of regular shape. This may be due to the settlement of bricks when fuel near bottom is burnt and turned to ashes.
- Very slow process.
- It isn't possible to regulate fire in a clamp. Once it starts burning and the bricks are liable to uneven burning.
- Quality of bricks is not uniform.
- Bricks near the bottom are overburnt and those near sides & top are underburnt.

(ii) Kilns: —
 → Intermittent
 → Continuous

- A kiln is a large oven used to burn bricks.
- Bricks prepared by kilns are better than clamps.

(a) Intermittent kilns:

- These kilns are intermittent in operation which means that they are loaded, fired, cooled and unloaded.
- These be either rectangular or circular in plan.
- They can be overground (or) underground.

→ Intermittent kilns are further divided into 2 groups.

(a) Intermittent up-draught kilns → not uniform →

Bottom = overburnt

Top = underburnt

(b) Intermittent down-draught kilns → Evenly burnt bricks.

→ These kilns are in the form of rectangular structure with thick outside walls.

→ Wide doors are provided at each end for loading & unloading of kilns.

→ Fuel channels (or) passages are provided to carry flames (or) hot gases through the body of kiln.

→ After loading the kiln, it is fired, loaded and unloaded for next loading.

→ Since wall and sides gets cooled during reloading and are to be heated again during next firing, there is a wastage of fuel.

→ Int. up-draught gives non-uniform burning bricks, bricks near bottom are overburnt & those near top are underburnt. Int. down draught kiln gives evenly burnt brick.

continuous kiln:

→ These kilns are continuous in operation i.e. loading, firing, cooling and unloading are carried out simultaneously in these kilns.

3 types of continuous kilns: ① Bull's trench kiln

② Hobman's kiln

③ Tunnel kiln

(i) Bull's trench kiln:

→ may be rectangular, circular or oval shape in plan.

As the name suggests, the kiln is constructed in a trench excavated in ground. It may be fully underground or partly projecting above ground.

* This is the most widely used kiln in India and it gives continuous supply of bricks.

(ii) Hoffman's kiln:

→ The kiln is constructed overground and hence it is called flame kiln.

→ Its shape is circular in plan & it is divided into a no. of compartments or chambers. As a permanent roof is provided, the kiln can even function during rainy season.

Qualities of good building bricks:-

Good building bricks have the following main qualities.

- ① Colour ⇒ Colour should be uniform and bright in colour.
- ② Shape ⇒ Brick should have plane faces. They should possess sharp and truly right angled corners.
- ③ Texture ⇒ Bricks should be fine, dense and uniform in texture. A fractured surface should not show fissures, cavities, loose grit and unburnt lime.
- ④ Size ⇒ Bricks should be of standard size prescribed by the code.
- ⑤ Sound ⇒ Bricks should give out a metallic sound when struck with a hammer (or) with another brick.

- ⑥ Hardness \Rightarrow Scratching with fingernail should not leave any impression on the brick.
- ⑦ Strength \Rightarrow Crushing strength should be as high as possible.
- ⑧ Water absorption \Rightarrow After immersing the bricks in cold water for 24 hrs W.A should not be more than 20%.
- ⑨ Efflorescence \Rightarrow Bricks should not show white patches when soaked in water for 24 hrs. White patches indicate the presence of sulphates of calcium (CaSO_4), Mg, Na, K (i.e. all types of salt). They keep the masonry permanently damp.
- ⑩ Thermal conductivity \Rightarrow Bricks should have low thermal conductivity so that building built with them are cool in summer and warm in winter.
- ⑪ Sound insulation \Rightarrow Light weight bricks and hollow

Strength $\propto \frac{1}{\text{Sound insulation}}$
 bricks provide good sound insulation.
 So bricks with unnecessarily high strength (heavy bricks) need not be used.

- ⑫ Fire resistance \Rightarrow Good bricks are fire resistant.

Cement by mason Joseph Aspdin of Leeds in England is 1824.

Natural cement is obtained by burning and crushing the stones containing clay, carbonate of lime, and some amount of carbonate of magnesia.

* India's per-capita cement production is 210kg per annum as of sep 2017.

* cement is a material which has cohesive and adhesive properties in the presence of water.

* For civil Engg work, i.e. construction industry its primary function is to bind the fine aggregates (sand) and coarse aggregate particles together into a hard compact durable mass.

Cement and lime:

→ cement can be used under conditions and circumstances which are not favourable for lime.

→ The cement, when converted into a paste form, sets quickly.

→ colour of cement and lime are different.

→ when water is added to the cement, no heat is produced and there is no slaking action.

Note:

* Cement is a product obtained by pulverizing clinker formed by calcinating raw materials primarily consisting of lime (CaO), silicate (SiO_2), Alumina (Al_2O_3) and iron oxide (Fe_2O_3).

Pulverize \Rightarrow make into a powder form or dust form

* Calcination \Rightarrow The process of heating a substance to a high temperature but below the melting or

Boiling point, causing loss of moisture, reduction on oxidation and dissociation into simple substances.

* cements used in the construction industry can be classified as hydraulic and non-hydraulic.

→ Hydraulic cement set and harden in water and give a product which is stable e.g. port land cement.

→ Non-hydraulic cement doesn't set and harden in water such as non-hydraulic lime (or) which are unstable in water e.g. Plaster of Paris.

* cement can be manufactured either from natural cement stones (or) artificially by using calcareous and argillaceous materials.

Examples of natural cements are Roman cement, Puzzolona cement and Medina cement.

Ex: of artificial cements are OPC, special cements —

Note:

Argillaceous ⇒ shale & clay, cement rock, Blast Furnace slag, marl.
↓
clay

Calcareous ⇒ Limestone, chalk, Marine shells

↓
calcium carbonate

* There are various varieties of artificial cements are available in the market at present; we will concentrate on OPC (or) normal setting (or) ordinary cement which has a production of about $\frac{2}{3}$ rd of the total production of cement.

Portland cement:

→ Most common variety of artificial ^{cement} and it is also known as normal setting cement (or) ordinary cement.

→ This cement is called as portland cement, because it has resemblance in its colour after setting to a variety of sand stone which is found in abundance in portland England.

1. Physical properties of cement:

→ Mainly depend upon its chemical composition, thoroughness of burning and fineness of grinding.

→ gives strength to masonry → an excellent binding material → easily workable.

→ offers good resistant to the moisture → possesses good plasticity.

→ hardens early ⇒ cement thrown in water should sink and shouldn't float on the surface.

→ Fineness, consistency, initial setting time, final setting time, soundness.

2. Mechanical properties of cement:

★ compressive strength @ end of 3 days & 11.5 MPa
@ end of 7 days & 17.5 MPa

★ Tensile strength @ end of 3 days & 2 MPa
@ end of 7 days & 2.5 MPa

3. Chemical properties of cement:

→ $\frac{Al_2O_3}{Fe_2O_3}$ (in%) & 0.66 ($\frac{2}{3}$).

→ $\frac{CaO}{Al_2O_3, Fe_2O_3, SiO_2}$ (%) = LSF (Lime Saturation Factor)

⇒ & 0.66
& 1.02

i.e. It should lie betⁿ

0.66 - 1.02

- Total loss on ignition $\neq 4\%$
- Total sulphur content $\neq 2.75\%$
- wt. of insoluble residue $\neq 1.5\%$
- wt. of magnesia $\neq 5\%$

As per IS: 10262 has classified the OPC, grade wise from A to F based on 28 days compressive strength as follows:

category	A	B	C	D	E	F
Strength (MPa)	32.5-37.5	37.5-42.5	42.5-47.5	47.5-52.5	52.5-57.5	57.5-62.5

↓ 33 grades
 ↓ 43 grade
 ↓ 53 grade But 43 grade

cement available in market Ball in category 'D' and that 53 grade cements in category F.

composition of ordinary cement:

2 basic ingredients → Argillaceous ⇒ clay
 Predominates
 → calcareous ⇒ $CaCO_3$
 Predominates

Ingredients

% composition

Functions

① Lime (CaO) → 62 → control strength & soundness & its deficiency reduce strength time so set quickly & its excess make cement unsound so cause expand & disintegrate. Excess of it causes slow setting.

② Silica $\rightarrow 22 \rightarrow$ Gives strength (Formation of C_2S, C_3S)

③ Alumina $\rightarrow 5 \rightarrow$ Responsible for quick setting. If, in excess, it lowers the strength.

④ Iron oxide $(Fe_2O_3) \rightarrow 3 \rightarrow$ Gives colour, hardness and strength to the cement and helps in fusion of different ingredients.

⑤ Magnesia $(MgO) \rightarrow 2 \rightarrow$ Gives colour, hardness to the cement. If it is in excess, causes cracks in mortar & concrete and unsoundness.

⑥ Calcium Sulphate $(CaSO_4) \rightarrow 3 \rightarrow$ It is in the form of gypsum and its function is to increase the initial setting time of cement (i.e. delayed the setting).

⑦ Sulphur $(SO_3) \rightarrow 1 \rightarrow$ Useful for making the cement sound. If in excess, it causes cement to become unsound.

⑧ Alkalies $\left[\begin{array}{l} K_2O + Na_2O \\ TiO_2, P_2O_5 \end{array} \right] \rightarrow 0.2-1 \rightarrow$ These are residues and if in excess cause efflorescence & cracking.

Note:

\rightarrow An increase in lime content beyond a certain value makes it difficult to combine completely with other compounds.

consequently, free lime will exist in the clinker and will result in an unsound cement.

\rightarrow An increase in silica content at the expense of alumina and ferric oxide makes the cement difficult to fuse and form clinker.

* (i) Alkali oxides K_2O and Na_2O (ii) Magnesia (MgO), are two oxides adversely affects the quality of cement, so treated as harmful constituents of cement.

Note:

- Rate of setting of cement paste is controlled by regulating the ratio of $\frac{SiO_2}{(Al_2O_3 + Fe_2O_3)}$.
- When development of heat of hydration is undesirable, the silica content is increased to about 21% and the alumina and iron oxide (Al_2O_3) (Fe_2O_3) contents are limited to 6% each.
- Resistance to the action of sulphate waters is increased by raising further the silica content to 24% and reducing the alumina & iron contents in 4% each.
- The variation in composition depends largely on the ratio of $\frac{CaO}{SiO_2}$ in the raw materials.
- Alkalis accelerate the setting of cement paste.

Composition of cement clinker:

- When the raw materials are put in kiln then is Buses and following compounds are formed and they are known as Bogue's compound.

Principal mineral compounds in OPC	Chemical Formula	Name	composition	Formation in Kiln	Function	Heat of hydration
1. Tricalcium silicate (C ₃ S)	3CaO.SiO ₂	Alite	50-60%	IV	Early Strength	500 J/g
2. Dicalcium silicate (C ₂ S)	2CaO.SiO ₂	Belite	25-30%	III	Later Strength (Final or ultimate strength of cement)	260 J/g
3. Tricalcium aluminate (C ₃ A)	3CaO.Al ₂ O ₃	celite	8-12%	II	Initial setting (Flash Set)	865 J/g
4. Tetracalcium aluminoferrite (C ₄ AF)	4CaO.Al ₂ O ₃ .Fe ₂ O ₃	Felite	6-8%	I	(Flash set)	420 J/g

$$1 \text{ cal} = 4.184 \text{ J}$$

quality ↑
 ↓ C₃S of cement

C₃S :-

- It is supposed as the best cementing material and is well burnt cement.
- It enables clinker easy to grind; increases resistance to breezing & throwing.
- It hydrates rapidly generating high heat and develops an early hardness and strength.
- Raising of C₃S content beyond the specified limits increases heat of hydration and solubility of cement in water.
- The hydrolysis of C₃S is mainly responsible for 7 days strength & hardness.
- Rate of hydrolysis of C₃S and the character of gel developed are the main cause of the hardness and early strength of cement.

C₂S:

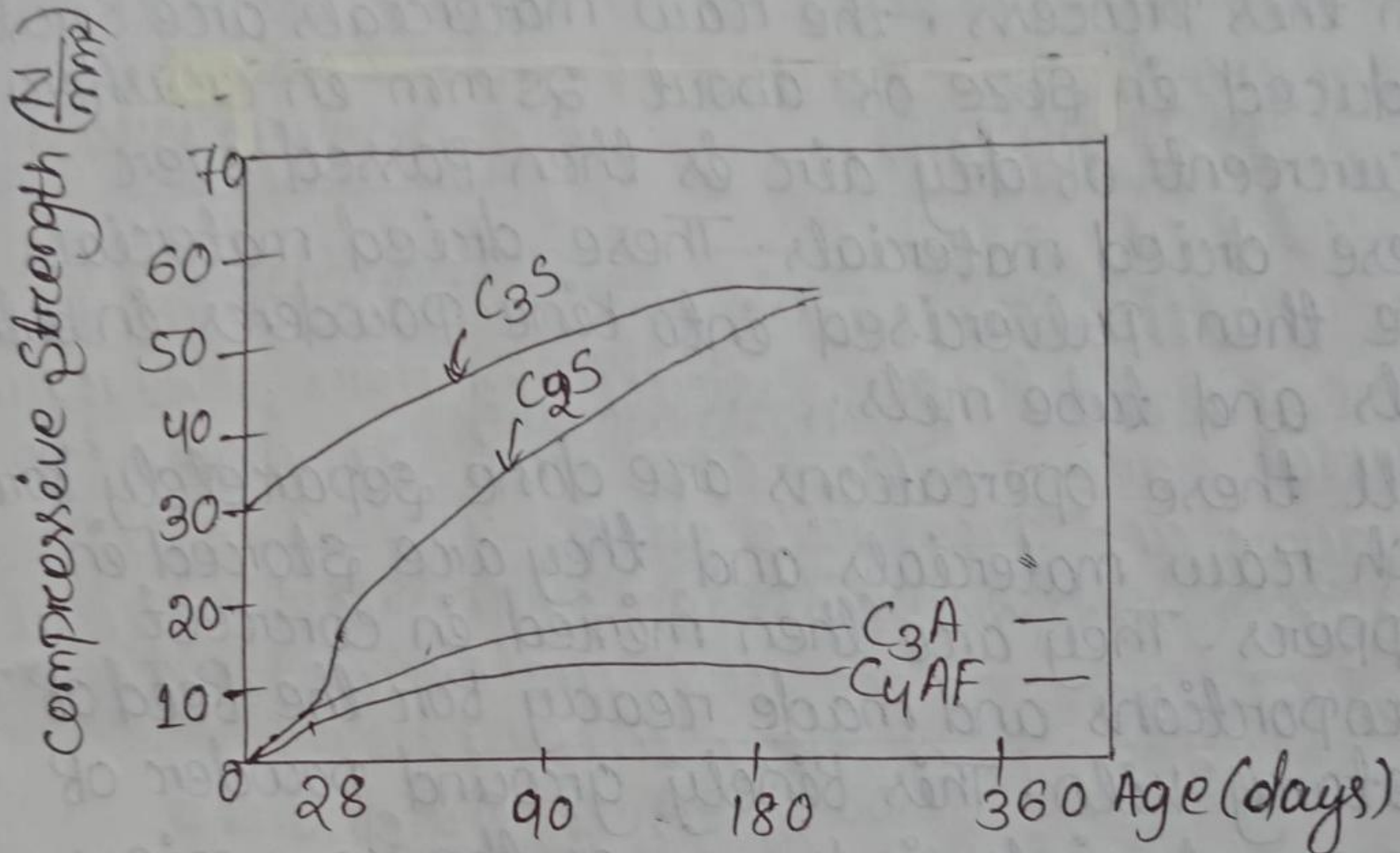
- It rapidly hydrates and hardens slowly and takes long time to add to the strength (after a year (or) more). ⇒ It imparts resistance to chemical attack.
- Excess (or) raising of C₂S content, renders clinker harder to grind, reduces early strength, decreases resistance to freezing and thawing at early ages and decreases heat of hydration.
- The hydrolysis of C₂S proceeds slowly.
- At early ages, less than a month, C₂S has little influence on strength and hardness. While after 1 year, its contribution to the strength and hardness is proportionality almost equal to C₃S.

C₃A:

- It rapidly reacts with water and is responsible for flash set of finely ground clinker.
- Rapidity of action is regulated by the addition of 2-3% gypsum at the time of grinding cement.
- It is responsible for the initial set, high heat of hydration and has greater tendency to volume changes causing cracking.
- Excess (or) raising the C₃A content reduces the setting time, weakens resistance to sulphate attack and lowers the ultimate strength, heat of hydration and contraction during air hardening.

C₄AF

- It is responsible for Blash set but generates less heat.
- It has poorest cementing value.
- Raising the C₄AF content reduces the strength slightly.



(Contribution of cement compounds to strength of cement)

v.v.g

Manufacturing of ordinary cement:

→ 3 distinct operations are involved in the manufacture of OPC.

1. Mining of materials
2. Burning
3. Grinding

1. Mining of materials:

raw material

- Argillaceous → Shale, Clay, Cement rock, Blast Furnace Slag, marl.
- Calcareous → limestone, Chalk, marine shells.

★ These raw materials can be mined either in dry condition (or) in wet condition. The process is accordingly known as dry process or the wet process of mixing.

Modern technology

↳ old technology.

(e) Dry process (or) Modern technology:

→ In this process, the raw materials are first reduced in size of about 25 mm in crushers. A current of dry air is then passed over these dried materials. These dried materials are then pulverised into fine powders in ball mills and tube mills.

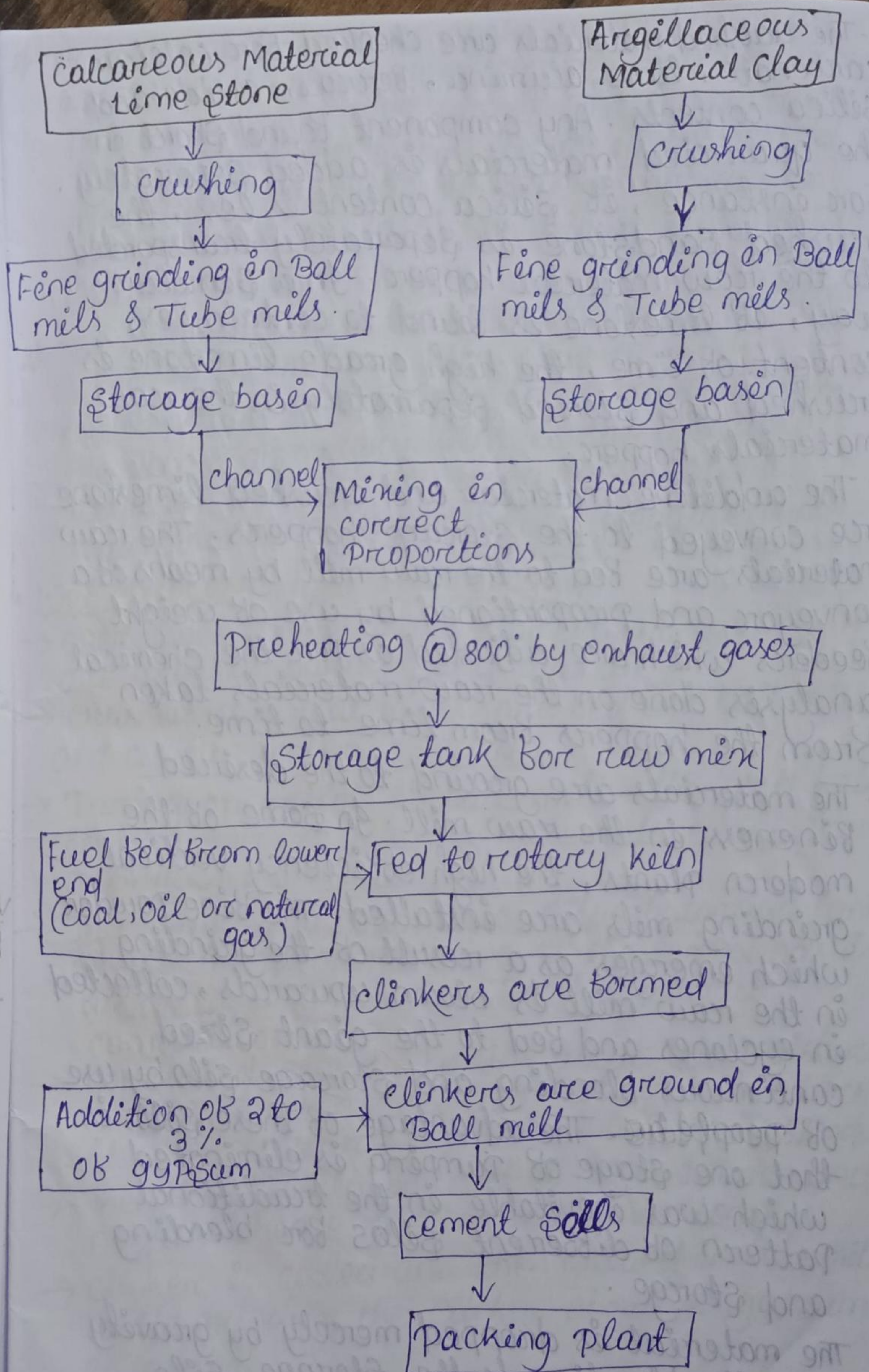
→ All these operations are done separately for each raw materials and they are stored in hoppers. They are then mined in correct proportions and made ready for the feed of rotary kiln. This finely ground powder of raw materials is known as the raw meal; and it is stored in storage tank.

Following is the procedure of manufacture of cement by the dry process / modern technology.

→ Most of the cement factories are located very close to the lime stone quarries. The boulders upto 1.2 m. size are transported in huge dumpers upto 300 k.N. capacity & dumped into the hopper of the crusher.

After crushing of lime stone about 75 cm size is moved from crusher by a series of conveyors for stacking.

→ The argillaceous (or) clay materials found in the quarry are also dumped into the crusher and stocked along with the limestone.



Flow diagram of dry process.

★ The crushed materials are checked for calcium carbonate, lime, alumina, ferrous oxide and silica contents. Any component found short in the quarried materials is added separately. For instance, if silica content is less, the crushed sandstone is separately transported to the raw material hopper. In a similar way, if limestone is found to contain less content of lime, the high grade limestone is crushed and stored separately in the raw materials hopper.

★ The additive material and crushed limestone are conveyed to the storage hoppers. The raw materials are fed to the raw mill by means of a conveyor and proportioned by use of weigh feeders which are adjusted as per the chemical analysis done on the raw materials taken from the hoppers from time to time.

★ The materials are ground to the desired fineness in the raw mill. In some of the modern plants, the high efficiency vertical grinding mills are installed. The fine powder which emerges as a result of the grinding in the raw mill is blown upwards, collected in cyclones and fed to the giant sized continuous blending and storage silo by use of aeropole. The advantage of these silos is that one stage of pumping is eliminated which was inevitable in the traditional pattern of different silos for blending and storage.

★ The material is dropped merely by gravity from the blending to the storage silo thereby conserving power.

★ The material is then once again pumped using an aeropole into the preheater. The most modern preheaters have five stages. The temperature of the material bed from the top is increased in stages from 60°C to 850°C as hot gas at temperature of 1000°C is blown against the falling granulant.

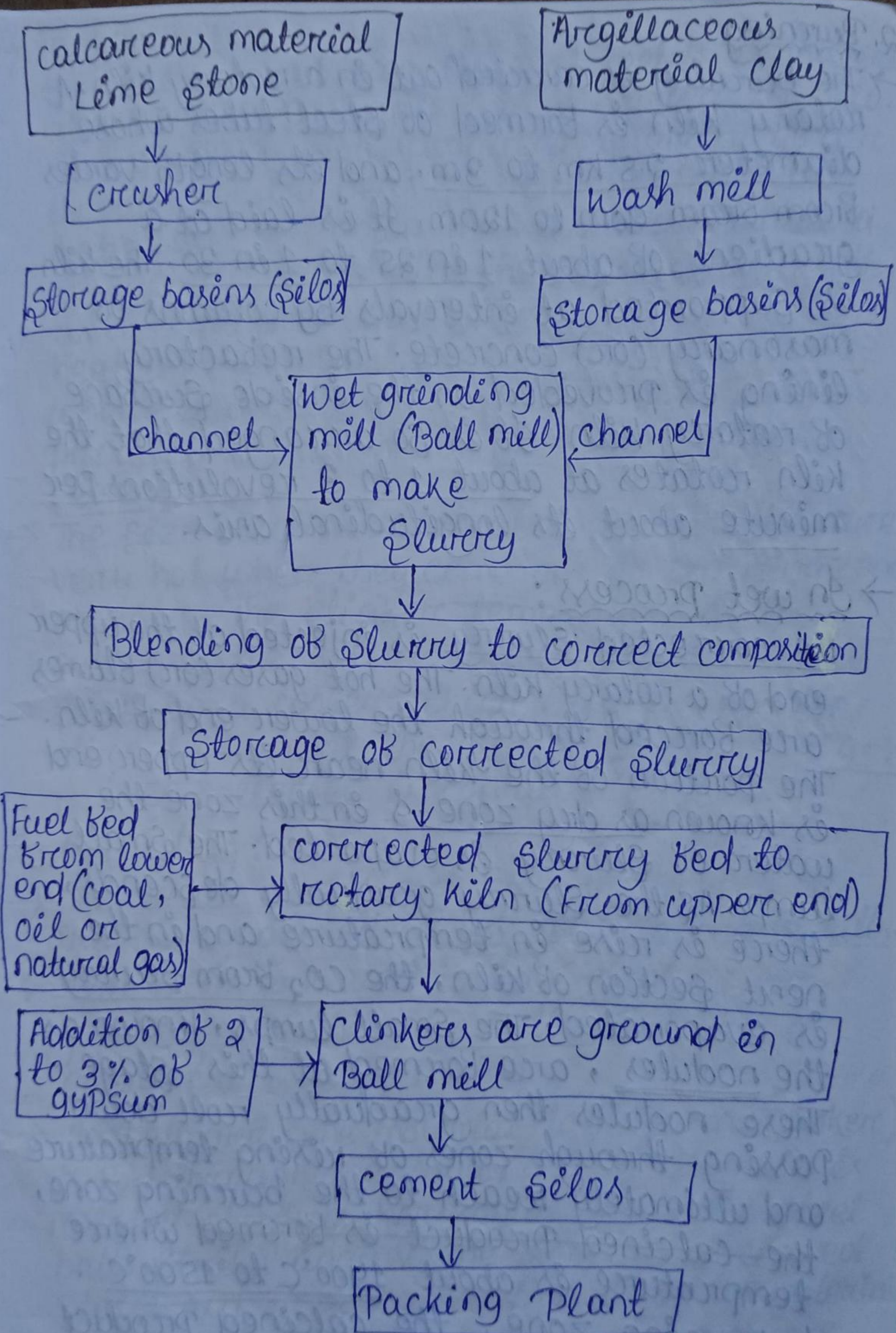
★ The material from the bottom of the preheater is fed to the rotary kiln. Due to the use of multi-stage preheaters in the modern plants, the length of rotary kilns is considerably reduced thereby resulting in saving of maintenance cost and power requirements.

Wet process : [old technology]

- operations in wet process of cement manufacture are mining, burning and grinding.
- Crushed raw materials are fed into ball mill and a little water is added.
- During operation of ball mill, the steel ball in it pulverize the raw materials which form a slurry with water.
- This slurry is passed to silos (storage tank), where the proportioning of the compounds is adjusted to ensure desired chemical composition.
- corrected slurry having about 40% moisture content, is then fed into rotary kiln where it uses moisture and forms into lumps/nodules.
- Then it is finally burned at $1500-1600^{\circ}\text{C}$ and nodules change to clinker at this temp.
- clinker is cooled and then ground in tube mills. While grinding the clinker, about 3% gypsum is added.
- The cement is then stored in silos from where it is supplied.

Comparison of wet and dry process :-

- Dry process is considered to be economical as compared to wet process because of less consumption of fuel in the kiln.
- Advantages of wet process are the low cost of excavating and grinding raw materials, the accurate control of composition and homogeneity of the slurry.
- But longer kilns are reqd in wet process which are more costly and less responsive to a variable clinker demand than the short kiln which can be used in dry process.



Flow diagram of Wet process.

2. Burning:

→ The burning is carried out in a rotary kiln. A rotary kiln is formed of steel tubes whose diameter 2.5 km to 3m. and its length varies from 90m to 120m. It is laid at a gradient of about 1 in 25 to 1 in 30. The kiln is supported at intervals by columns of masonry (or) concrete. The refractory lining is provided on the inside surface of rotary kiln. It is so arranged that the kiln rotates at about 1 to 3 revolutions per minute about its longitudinal axis.

→ In wet process:

the corrected slurry is injected at the upper end of a rotary kiln. The hot gases (or) flames are forced through the lower end of kiln. The portion of the kiln near its upper end is known as dry zone. & in this zone the water of slurry is evaporated.

As the slurry gradually descends, there is rise in temperature and in the next section of kiln, the CO₂ from slurry is evaporated. The small lumps, known as the nodules, are formed at this stage. These nodules then gradually roll down passing through zones of rising temperature and ultimately reach to the burning zone, where temperature is about 1400°C to 1500°C.

In burning zone, the calcined product is formed and nodules are converted into small hard dark greenish blue balls which are known as clinkers.

→ In dry process:

the coal brought from the coal fields is pulverised in vertical coal mill & it is stored in silo. It is pumped with required quantity of air through the burners. The pre-heated raw materials roll down the kiln and get heated to such an extent that the CO_2 is driven off with combustion gases. The material is then heated to temperature of 1400°C to 1500°C when, it gets fused together. The fused product is known as clinker (or) raw cement.

→ The size of clinker about 3-20 mm & they are very hot when they come out of burning zone of kiln. The clinker temperature at the outlet of the kiln is nearly 1000°C . There shall be a cooler which laid in opposite direction, cool the clinker to temp. of 95°C & then collected in containers.

3. Grinding:

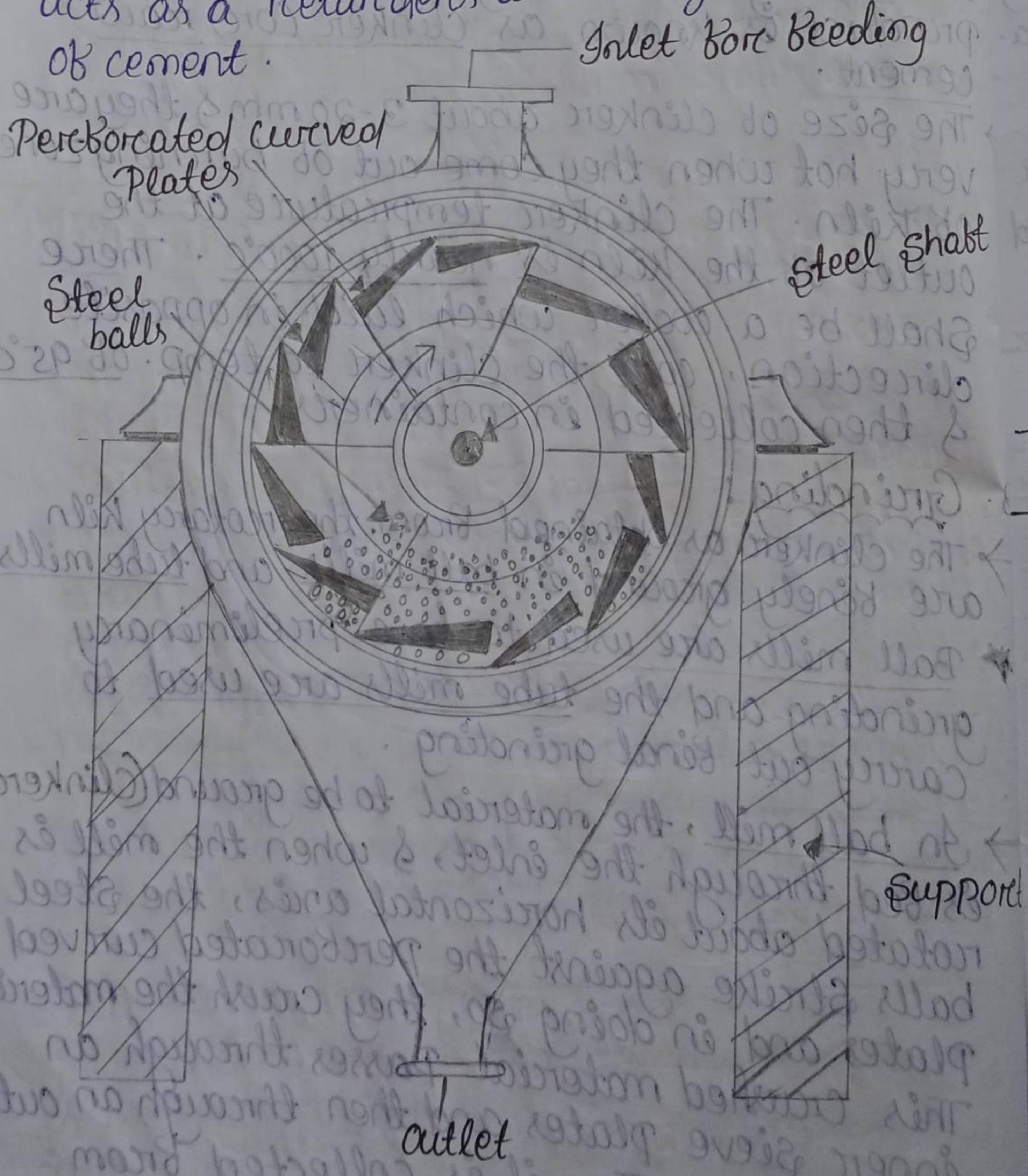
→ The clinker as obtained from the rotary kiln are finely ground in ball mills and tube mills.

★ Ball mills are used to have preliminary grinding and the tube mills are used to carry out final grinding.

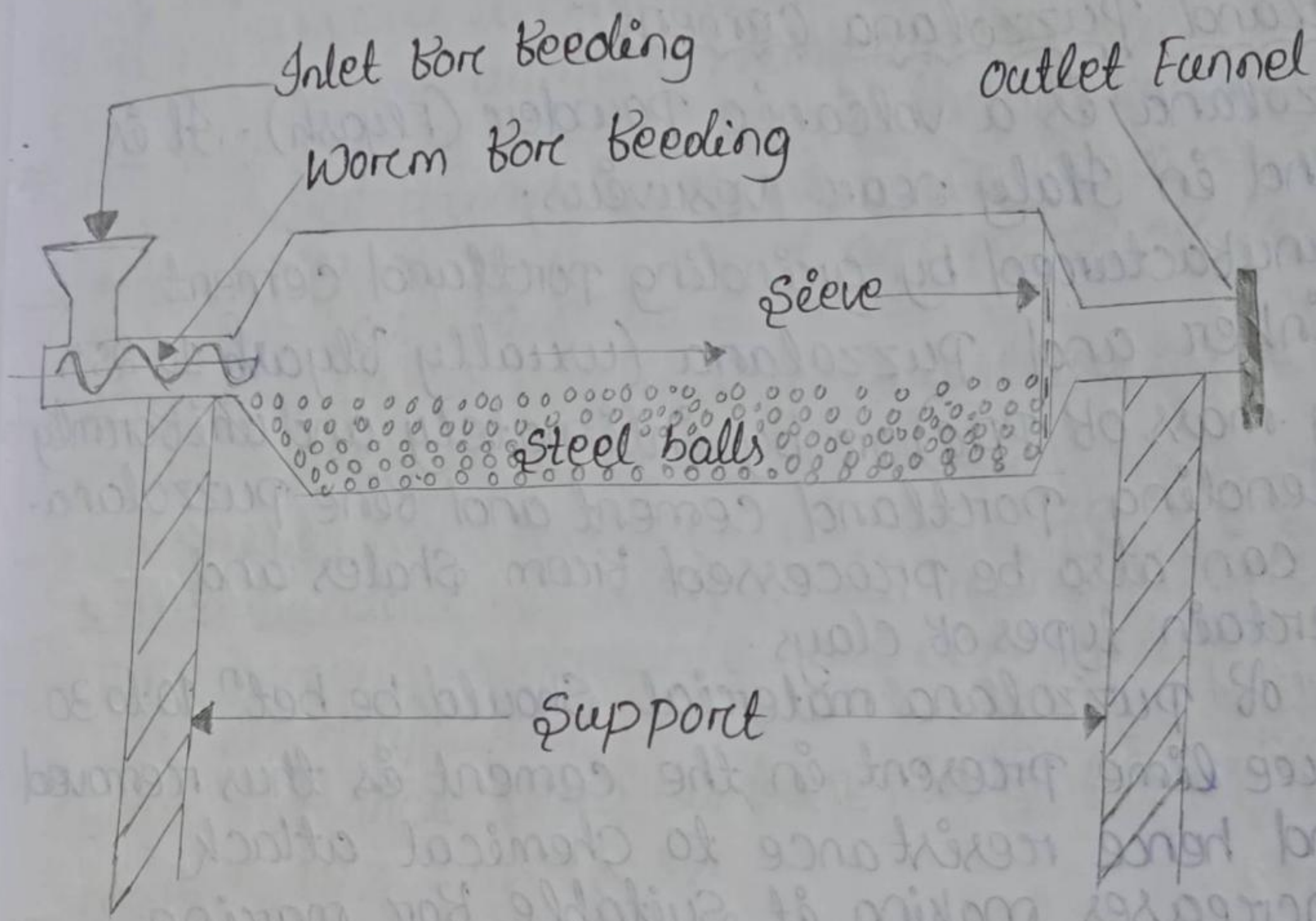
→ In ball mill, the material to be ground (clinker) is fed through the inlet, & when the mill is rotated about its horizontal axis, the steel balls strike against the perforated curved plates and in doing so, they crush the material. This crushed material passes through an inner sieve plate and then through an outer sieve plate. Then it is collected from an outlet.

→ In tube mill, fine grinding is achieved due to steel balls of smaller size. A worm is provided to feed the material to the mill. The pulverised material is collected at the outlet funnel.

★ During the grinding a small quantity about 3 to 4% of gypsum is added. Gypsum controls initial setting time of cement. So gypsum acts as a retarder and delays the action of cement.



Ball mill for grinding cylinders



Tube mill Ball grinding cylinders

Types of cement :

→ There are various types of cement available in the market. And the different special types of cements are manufactured by altering the chemical composition of OPC, by using additives (or) by using different raw material.

1. PPC (Portland Pozzolana cement)
2. PSC (Portland Slag cement)
3. RHC (Rapid Hardening cement)
4. HAC (High Alumina cement)
5. SSC (Super Sulphated Portland cement)
6. SRC (Sulphate Resisting Portland cement)
7. LHC (Low heat cement)
8. Quick setting cement
9. White and coloured cement

- 10. Hydrophobic cement
- 11. Air entraining cement

Portland Pozzolana Cement:

Pozzolana is a volcanic powder (Flyash). It is found in Italy near Vesuvius.

- Manufactured by grinding portland cement clinker and pozzolana (usually flyash 10-25% by mass of PPC) or by intimately and uniformly blending portland cement and fine pozzolana
- It can also be processed from shales and certain types of clays.
- % of pozzolana material should be betⁿ 10 to 30.
- Free lime present in the cement is thus removed and hence resistance to chemical attack increases making it suitable for marine works.

Properties:

- It has low rate of development of strength but ultimate strength is comparable with OPC.
- Addition of pozzolana to cement increases later strength and it offers greater resistance to the attack of aggressive water.

compressive strength	72 + 1hr ⇒ 16 MPa
	168 + 2hr ⇒ 22 MPa
	672 + 4hr ⇒ 33 MPa

Setting times are same as of OPC.

- Drying Shrinkage → 0.15% → Fineness → $300 \frac{m^2}{kg}$

Uses: It has low heat evolution and is used in the places of mass concrete such as dams, and in places of high temperature.

Portland Slag cement: (PSC) (slag is a non-metallic product)
Slag is a waste product in the manufacturing process of pig-iron and it contains the basic elements of cement, namely alumina, lime and silica.

* For this cement slag is used & the clinkers of cement are ground with about 60-65% of slag.

→ properties of these cement are more or less the same as those of OPC.

→ Its strength in early days is less and hence it requires longer curing period.

→ It proves to be economical, as slag which is a waste product, is used in its manufacture.

→ This cement is durable, but not suitable for use in dry arid zones.

Blast furnace slag contains $\text{CaO} \approx 45\%$, $\text{SiO}_2 \approx 35\%$,
 $\text{Al}_2\text{O}_3 \approx 12\%$, $\text{MgO} \approx 8\%$

Grades of cement:

→ It is classified either on the basis of specific surface area (or) crushing strength of concrete.

Specific surface area (S.S.A.):

→ defined as total surface area of cement particle in unit mass. units: $\frac{\text{cm}^2}{\text{g}}$, $\frac{\text{m}^2}{\text{kg}}$

① 33 grade cement: Its SSA = $225 \frac{\text{m}^2}{\text{kg}}$

② 43 grade cement: Its S.S.A = $340 \frac{\text{m}^2}{\text{kg}}$ → Mostly

used in general construction.

③ 53 grade cement: Its S.S.A $\gg \gg 340 \frac{\text{m}^2}{\text{kg}}$

Crushing strength:

→ Defined as the max^m compressive stress, developed in a cube just before failure.

Mortar:

→ It is the mixture of B.M + F.A + water.

Lime + F.A + water ⇒ Lime mortar.

Cement + F.A + water ⇒ Cement mortar.

Cement + Lime + F.A + water ⇒ Gauged mortar

↓
[Cement + lime mortar]
(or)

[lime + cement mortar]

★ For 33 grade cement ⇒ 33 crushing strength & 33MPa.

★ For 43 grade cement ⇒ 43 crushing strength & 43MPa.

★ For 53 grade cement ⇒ 53 crushing strength & 53MPa.

Note:

→ 33 grade cement is outdated.

→ 43 grade cement ~~is~~ can be used for general construction.

→ 53 grade cement can be used in high strength concrete.

★ 53 grade cement can't be used directly for general construction because shrinkage can occur.

If it is compulsory to be use then add some klyash before use it, it will reduce the strength.

Mortar & concrete:

Sand:

Sand is a natural product obtained as sea sand, river sand, malla sand and pit sand. These days, manufactured sand is prepared by crushing harder varieties of stones. However, sea sand is not used for making mortar because of the following reasons:

- It contains salt and hence, the structure built with such mortar remains damp. After some time, blisters (white patches) appear on the surface.
- It contains shells and other organic matters, which decompose after some time, reducing the life of the mortar.

v.v.s.

Bulking of sand:

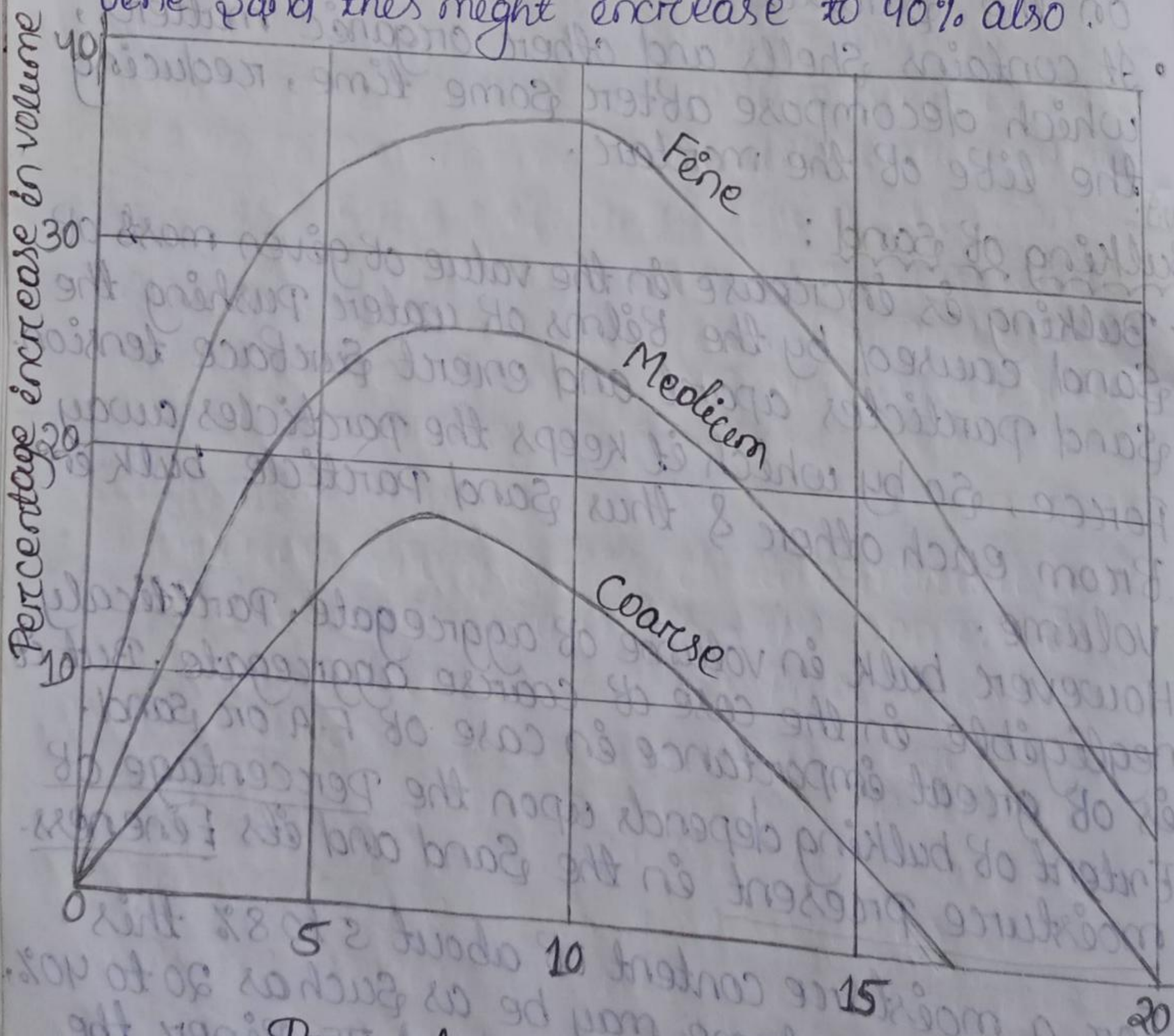
- Bulking is increase in the volume of given mass of sand caused by the films of water pushing the sand particles apart and exert surface tension force, so by which it keeps the particles away from each other & thus sand particle bulk in volume.
- However bulk in volume of aggregate, practically negligible in the case of coarse aggregate. But it is of great importance in case of F.A or sand.
- Extent of bulking depends upon the percentage of moisture present in the sand and its fineness.

For a moisture content of about 5 to 8% this increase of volume may be as such as 20 to 40%, depending upon grading of sand. The finer the material the more will be the increase in volume for a given moisture content.

→ So bulking increases with moisture content up to a certain point (4-6%) reaches a maximum, the film of water on the sand surface breaks, and then it starts decreasing till it finally at 25-30% of m.c, the volume of sand returns to its original volume when it is dry.

→ In preparing concrete mixes if sand is measured by volume and no allowance is made for bulking, the moist sand will occupy considerably larger volume than that prepared by the dry sand and consequently the mix will be richer.

→ In ordinary sand bulking varies from 15 to 30% for increase in moisture content 5 to 8%. For fine sand this might increase to 40% also.



Percentage by weight of moisture
Chart showing bulking of sand

The mortars are classified into the following five categories:

(i) Lime mortar

(ii) Surkhi mortar

(iii) Cement mortar

(iv) Gauged mortar

(v) Gypsum mortar

(i) Lime mortar:

In this type of mortar, the lime is used as binding material.

The lime may be fat lime or hydraulic lime.

The fat lime shrinks to a great extent and hence it requires about 2 to 3 times its volume of sand.

The lime should be slaked before use. This mortar is unsuitable for water-logged areas or in damp situations.

For hydraulic lime, the proportion of lime to sand by volume is about 1:2 or so. This mortar should be consumed within one hour after mixing. It possesses more strength and can be used in damp situations.

The lime mortar has a high plasticity and it can be placed easily. It possesses good cohesiveness with other surfaces and shrinks very little.

It is sufficiently durable, but it hardens slowly. It is generally used for lightly loaded above-ground parts of buildings.

(ii) Surkhi mortar:

This type of mortar is prepared by using bulley surkhi instead of sand or by replacing half of sand in case of fat lime mortar. The powder of surkhi should be fine enough to pass BIS No. 9 sieve and the residue should not be more than 10% by weight.

The surkhi mortar is used for ordinary masonry work of all kinds in foundation and

Superstructure. But it cannot be used for Plastering or Pointing since Surkhi is likely to disintegrate after some time.

(ii) cement mortar:

In this type of mortar, the cement is used as binding material. Depending upon the strength required and importance of work, the proportion of cement to sand by volume varies from 1:2 to 1:6 or more.

It should be noted that Surkhi and cinder are not chemically inert substances and hence they cannot be used as adulterants with matrix as cement. Thus the sand only can be used to form cement mortar.

The proportion of cement with respect to sand should be determined with due regard to the specified durability and working conditions.

The cement mortar is used where a mortar of high strength water-resisting properties is required such as underground construction, water saturated soils, etc.

Gauged mortar:

To improve the quality of lime mortar and to achieve early strength, the cement is sometimes added to it. This process is known as the gauging. It makes lime mortar

economical, strong and dense. The usual proportion of cement to lime by volume is about 1:6 to 1:8. It is also known as the

composite mortar or lime-cement mortar and it can also be formed by the combination of cement and clay. This mortar may be used for bedding and for thick brick walls.

(V) Gypsum mortar:

These mortars are prepared from gypsum binding materials such as building gypsum and anhydrite binding materials.

Preparation of concrete

M30 = 1:2:2

A. proportioning of concrete ingredient

→ The process of selection of relative proportions of cement, sand, CA & water so as to obtain a concrete of desired quantity is known as the proportioning concrete.

→ It is observed that if a vessel is taken and filled with stones, of equal size, the voids to the extent of about 45% are formed. This result is independent of the size of stones. It is interesting to note that if sand is taken in place of stones, the same result will be obtained. ⇒ The result can be verified by pouring water in the vessel till it is full. The volume of water added in the vessel represents the amount of voids.

on, → The theory of formation of concrete is based on this phenomena of formation of voids. When CA is placed, such voids are formed; when F.A i.e. sand is added, it occupies these voids. Further; when finely powdered cement is added, it occupies the voids of sand particles. During the process of setting, a chemical reaction takes place betⁿ water and cement. This results in an absolutely solid substance.

→ In general proportions of CA, F.A cement and water should be such that the resulting concrete has the following properties.

(i) when concrete is fresh, it should have enough workability so that it can be placed in the formwork economically.

- (ii) The concrete must possess max^m density i.e. it should be the strongest and most watertight.
- (iii) The cost of materials and labour req^d to form the concrete should be minimum.

Different methods of proportioning concrete:

1. Arbitrary method.
2. Fineness modulus method.
3. Minimum voids method.
4. Maximum density method.
5. Water - cement ratio method.

Water - cement ratio method:

According to Abram, if concrete is assumed to be fully compacted, then the lower water-cement produces stibb paste having greater binding property and hence the lowering of w/c ratio within certain limits results in increased strength.

→ Similarly the higher water content increases the workability. But it is not useful for the chemical action. The excess water evaporates leaving pores in the concrete. Thus the increased water-cement ratio lowers the strength of concrete.

(B) Measurement of materials:

(i) mass batching: 1 bag cement contains 50 k.g. mass of cement.

(ii) volume batching: volume of 50 kg cement = 34.7 lit.
For volume boxes of

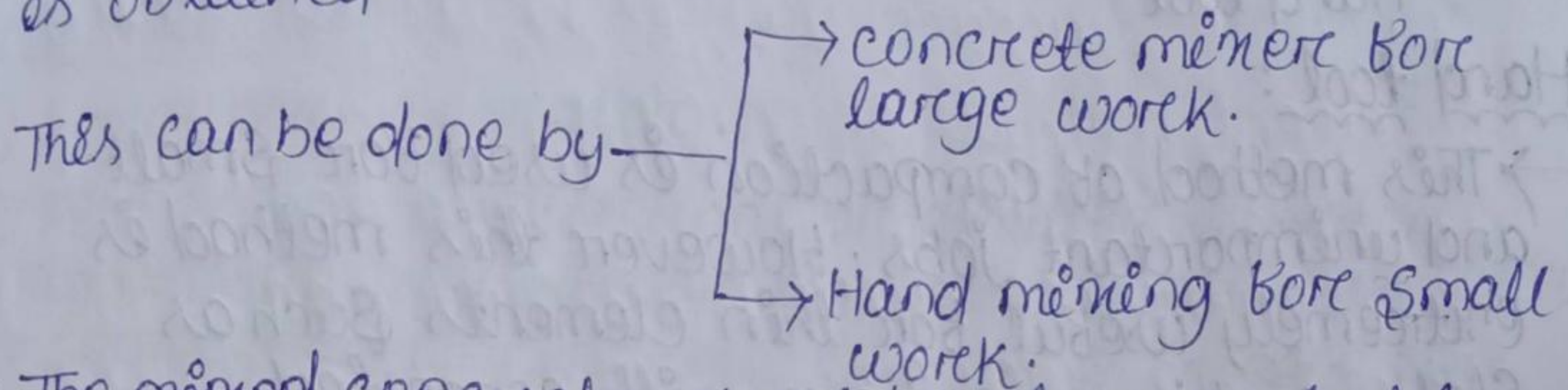
size (30x30x38) cm are used. size is in

Cubical shape $V = \frac{M}{\rho} = \frac{500 \text{ kg}}{1440 \frac{\text{kg}}{\text{m}^3}} = 0.0347 \text{ m}^3$
= 35 lit.

C. Mixing and placing of concrete:

→ The ingredients of concrete should be thoroughly mixed such that the cement paste is coated to the surface of all aggregates and a uniform mass is obtained.

This can be done by



→ The mixed concrete should be transported to the place of laying as early as possible.

→ During transport care should be taken to see that segregation doesn't take place and the concrete should be placed before it starts setting.

→ If segregation does occur during unloading the concrete should be remixed before placing.

D. compaction:

After concrete is placed at the desired location, the next step in the process of concrete production is its compaction.

compaction consolidates fresh concrete within the moulds or frameworks and around embedded parts and reinforcement steel.

→ compaction of the concrete is the process to get rid of the entrapped air and voids.

IMP → compaction is extremely important as 1% of voids can give a loss of 5% to 6% of strength, 5% of void can give a loss of 30% of strength, 10% of void can give a loss of 60% in strength, 25% of voids can give a loss of 90% in strength.

★ Other properties of concrete like durability, impermeability etc. also greatly depends on the compaction of concrete.

Compaction may be done by

Hand tool

Vibrator

Hand tool:

- This method of compaction is used for small and unimportant jobs. However this method is extremely useful for thin elements such as slabs and for members with congested reinforcements.
 - This method can be used for mixes with any workability except for very fluid or very plastic mix.
- Hand compaction is achieved by rodding, ramming or tamping.

Compaction by vibration:

- This is the most common and widely used method of compacting concrete for any structural element.
- The vibration imparted to the fresh concrete reduce the internal friction betⁿ the particles of concrete by setting the particles in motion and thus produce a dense and compact mass.
- On vibration, the concrete mix gets fluidize and the internal friction betⁿ the aggregate particles reduces, resulting in entrapped air to rise to the surface.
- On losing entrapped air the concrete gets denser.

The various types of vibrators in use are needle, formwork, table or platform and surface vibrator.

(i) Needle vibrator:

- These are also known as immersion, internal or poker vibrator.
- Needle vibrator can be used for any type of concrete work.

(ii) Formwork vibrator:

- Also known as external or shutter vibrator. Generally used under the following circumstances.
- compaction of concrete is reqd to be done in a very thin or very densely congested reinforced section.
- In addition to internal vibration, compaction is reqd to be done specially in the cover area where at times needle vibrator is unable to do satisfactory compaction.
- compaction of very stiff concrete is reqd to be done because such concrete can't be compacted by internal vibrator.

★ Formwork vibrators are used for concreting columns, thin walls and precast units. These are rigidly clamped to the formwork, causing it to vibrate and consequently transfer the vibrations to concrete.

(iii) Surface vibrators:

- These are also known as screed board vibrators. Surface vibrators are used for floor and roof slabs and pavement surfaces.
- ★ These are effective only upto a thickness of 150 mm (15 cm) of concrete but can be used upto 25 cm.
- Surface vibrators cause movement of fine particles to the top and hence aid the finishing operation. The operating frequency is 4000 cycles/min.

Note: compaction of concrete through vibrator is very useful. However over-vibration makes the concrete non-homogeneous.

E. Cure-ing:

→ The process by which the loss of water from concrete is prevented is known as cure-ing.
(or) The process by which keeping the concrete surface wet is known as curing.

Different methods of curing.

- (i) Moist curing. Spraying
- (ii) Membrane curing. Pondage
- (iii) Steam curing.

Segregation

Segregation

It is the separation of coarse aggregate from the concrete mass generally occurs because of non-reactive aggregate.

Bleeding

It is the separation of cement particles from the concrete mass in which cement floats on water.

Test on concrete

Test on Freshly mixed concrete

(To know the workability)

- Slump test
- Compaction Factor test
- Vee-bee test

Test on hardened concrete

Destructive test

- Comp. Strength
- Tensile Strength

Non-destructive test

- Rebound hammer test
- Pulse velocity test

Test on Freshly mixed concrete:

Workability :- Property of concrete

ability to work

- Workability is the ease with which a concrete can be mixed, placed and compacted so that a dense concrete is obtained (full compaction).
- In wide sense workability is defined as the amount of useful internal work necessary to produce full compaction.
- The workable concrete shouldn't show any segregation & bleeding of concrete.
- It should be noted that workability is different from consistency. consistency indicates fluidity or mobility.
- concrete with high consistency may not be workable for a particular job and concrete having same consistency may vary in workability.
- For example: concrete workable for foundation may not be workable for slab. Even for slab different workabilities will be required for compaction by hand and that by vibration. It is because the requirement of workability is less.

Segregation:

Segregation is said to occur when the constituent material of concrete try to separate out from each other, producing concentration of coarser material at one place and finer material at other place in concrete. Such concrete contains large voids and is less durable.

→ Segregation occurs due to poor grading of aggregate (i.e. large difference in size of particles) over vibration and dropping the concrete from above a certain height, which should be avoided & also due to concrete carried over long distance.

→ To reduce segregation, well graded aggregates are used and concrete is placed with enough compaction. The concrete should not be dropped from a height of more than 1.5 meters, increasing small size of C.A, air-entrainment, using dispersing agents and pozzolona.

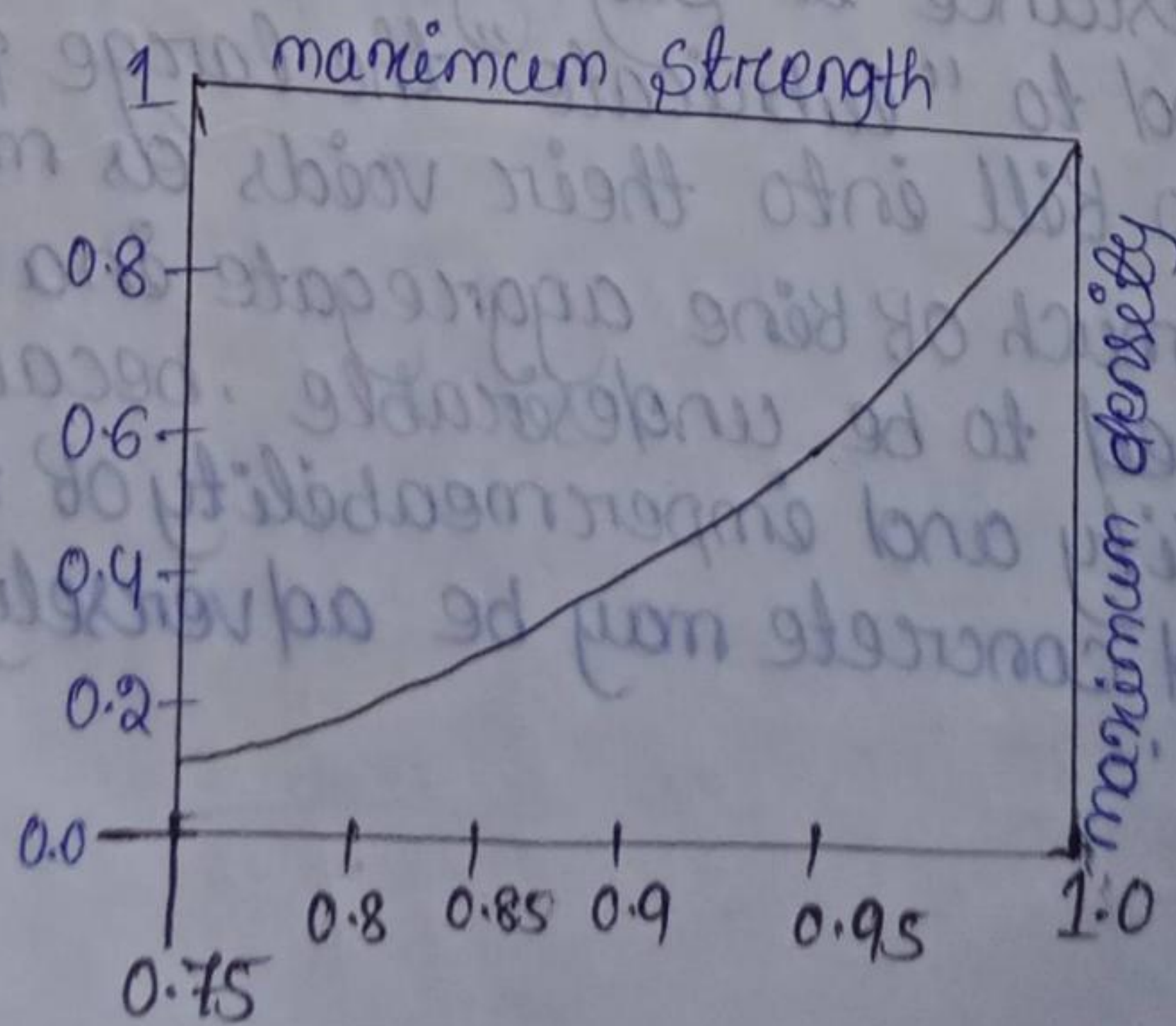
Bleeding:

Bleeding of concrete is said to occur when unreacted water in the mix tends to rise to the surface of freshly placed concrete due to sedimentation of constituents of concrete.

→ This produces continuous capillary pores which provides a clear straight access to chemicals and deleterious materials in concrete and lowers the strength & durability of concrete.

Grading requirements of aggregate:

- Grading is the size distribution of aggregate, it is measured by sieve analysis and is generally described by means of a grading curve, which shows the 'cumulative percentage passing' against the standard IS sieve sizes.
- The grading (as well as the type & size) of aggregate is a major factor which influences the workability of fresh concrete and its consequent degree of compaction.
- It is of extreme importance with regard to the quality of hardened concrete, because incomplete compaction results in voids, thereby lowering the density of concrete and preventing it from attaining its full compressive strength capability also the impermeability and durability characteristics get adversely affected.
- ★ It is observed that as little as 5% of voids can lower the strength by as much as 32%.



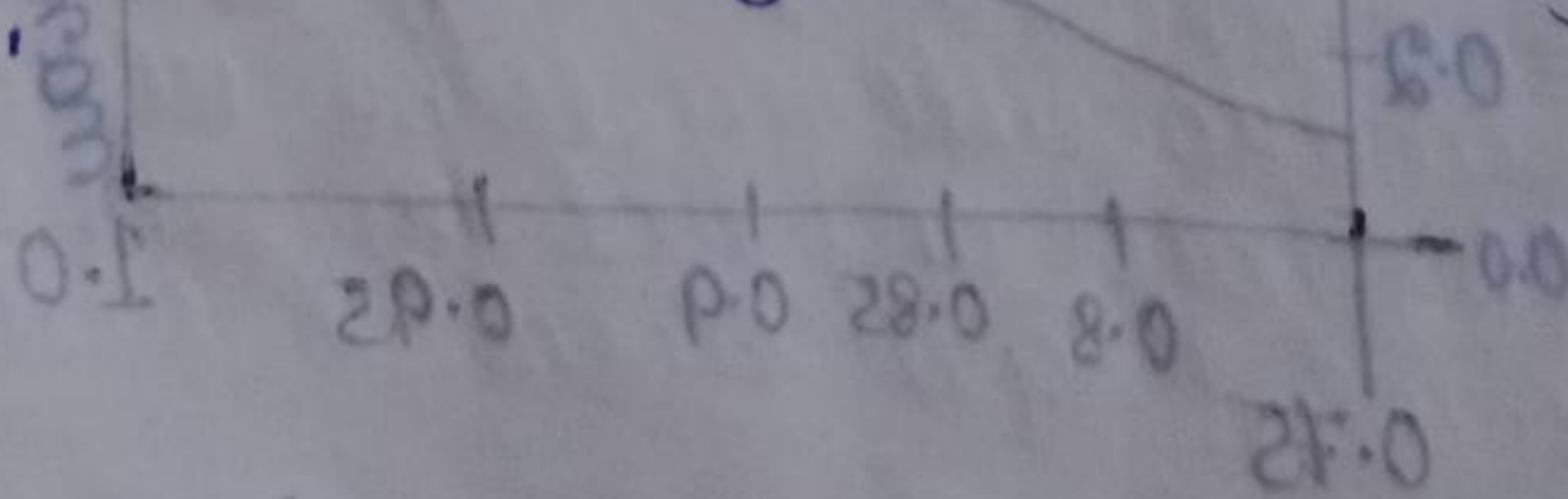
(Relⁿ betⁿ density ratio and strength ratio)

→ From an economic view point, it is desired to aim for maximum density by a proper grading of aggregate alone - with the smaller particles filling as much as possible, into the voids of the larger particles in the dry state, thereby limiting the use of the (more expensive) cement paste to filling in the voids in the fine aggregate. Unfortunately such a concrete mix is prone to be "harsh" and "unworkable". Moreover, it is very likely to segregate, with the coarse particles separating out on setting - more than the finer particles.

→ The cement paste must be in sufficient quantity to be able to coat properly all the aggregate surfaces, to achieve the reqd workability, and to ensure that the particles sizes are distributed as homogeneously as possible without segregation.

→ presence of more fines (sand & cement) in a mix is bound to improve both workability and resistance to segregation, because the fines tend to "lubricate" the large particles, and also fill into their voids as mortar.

→ But too much of fine aggregate in a mix is considered to be undesirable, because the durability and impermeability of the hardened concrete may be adversely affected.

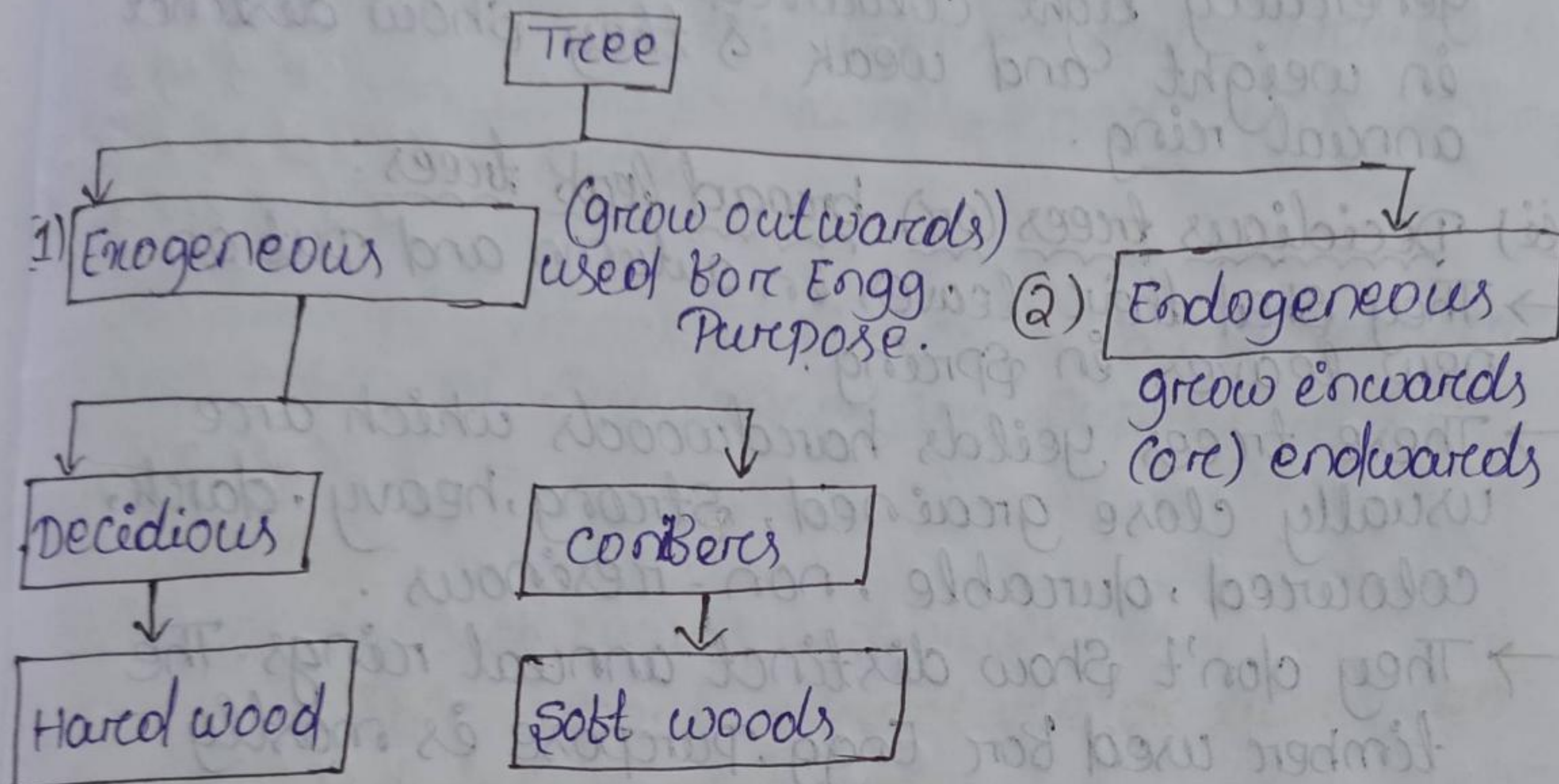


(Ref. per. of density ratio and specific gravity)

Timber

Classification of timber:

→ classified 2 types according to their manner of growth.



Endogeneous trees:

- grows inward or endwards.
- stems of these trees are too brittle, so not much suitable for Engg. Purpose.
- Found in tropical countries & limited application for temporary construction.

Ex: Cane, Bamboo, Palm etc.

Exogeneous trees:

- grows outwards & increase in bulk by the formation of successive annular rings on the outside under the bark.
 - These rings are known as annual rings because one such ring is added every year and these rings are useful in predicting the age of a tree.
 - Exogeneous trees are used for Engg. Purposes.
- Exogeneous trees are of 2 types.

- (e) conifers (or) evergreen trees:
- They remain evergreen and bear fruits in cones.
 - They have needle pointed leaves.
 - These trees yield softwoods which are generally light coloured, resinous, light in weight and weak & they show distinct annual ring.

- (ii) Deciduous trees (or) broad leaf trees:
- They shed their leaves in autumn and put on new leaves in spring.
 - These trees yield hardwoods which are usually close grained, strong, heavy, dark, coloured, durable, non-resinous.
 - They don't show distinct annual rings. The timber used for Engg. Purpose is mostly derived from deciduous trees.

Aspects	<u>Soft woods</u> conifers trees	<u>Hard woods</u> Deciduous trees.
1. Trees from which obtained.	conifers trees	Deciduous trees.
2. Annual rings	Distinct	Indistinct
3. Medullary rays	Indistinct	Distinct
4. Structure	Resinous and split easily	Nonresinous and close grained
5. Density	low	High
6. Fire resistance	Poor	More
7. Weight	Light	Heavy
8. Strength	Strong for direct pull and weak for resisting burst and shear also strong along the grain.	Equally strong for resisting tension, compression & shear. Also strong along and across the grain.
	Ex: Kail, Deodar, chir, Bir, pine, Spruce etc.	Ex: Babul, Mohogany, Sal, oak, teak etc. coconut.

Structure of timber:

⇒ Tree basically 3 parts:

1. Trunk (or) Stem ⇒

It support crown and supply water and nutrient from the roots to the leaves through branches and leaves through branches and from leaves back to the roots.

2. Roots ⇒

It implants trees in the soil, to absorb water and mineral substances and supply it to stem.

3. Crown ⇒

It consists of leaves and branches.

⇒ From visible aspect, structure of tree can be divided into 2 categories.

1. Macrostructure.

2. Microstructure.

1. Macrostructure can be study by cutting the trunk in 3 direction.

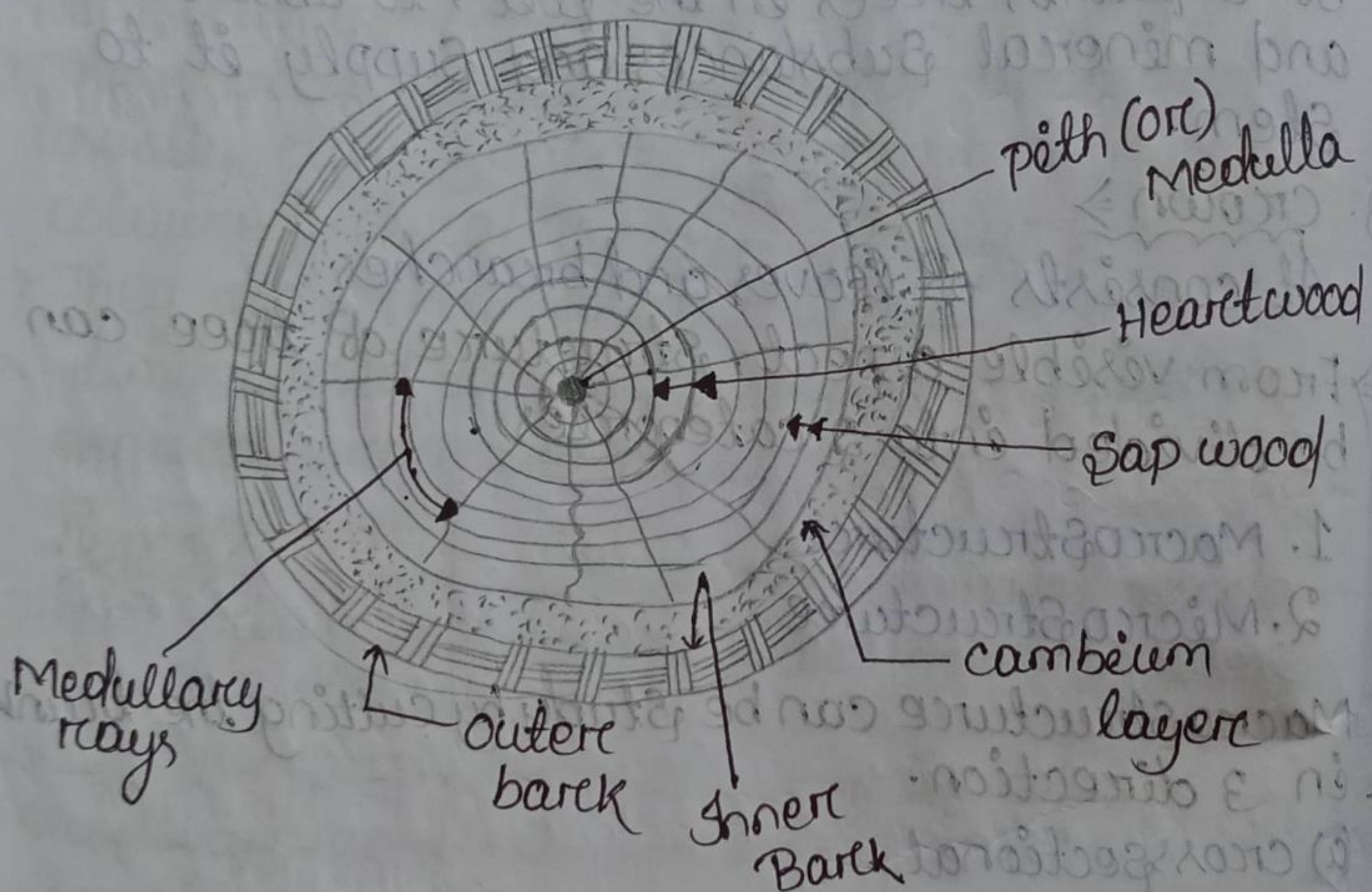
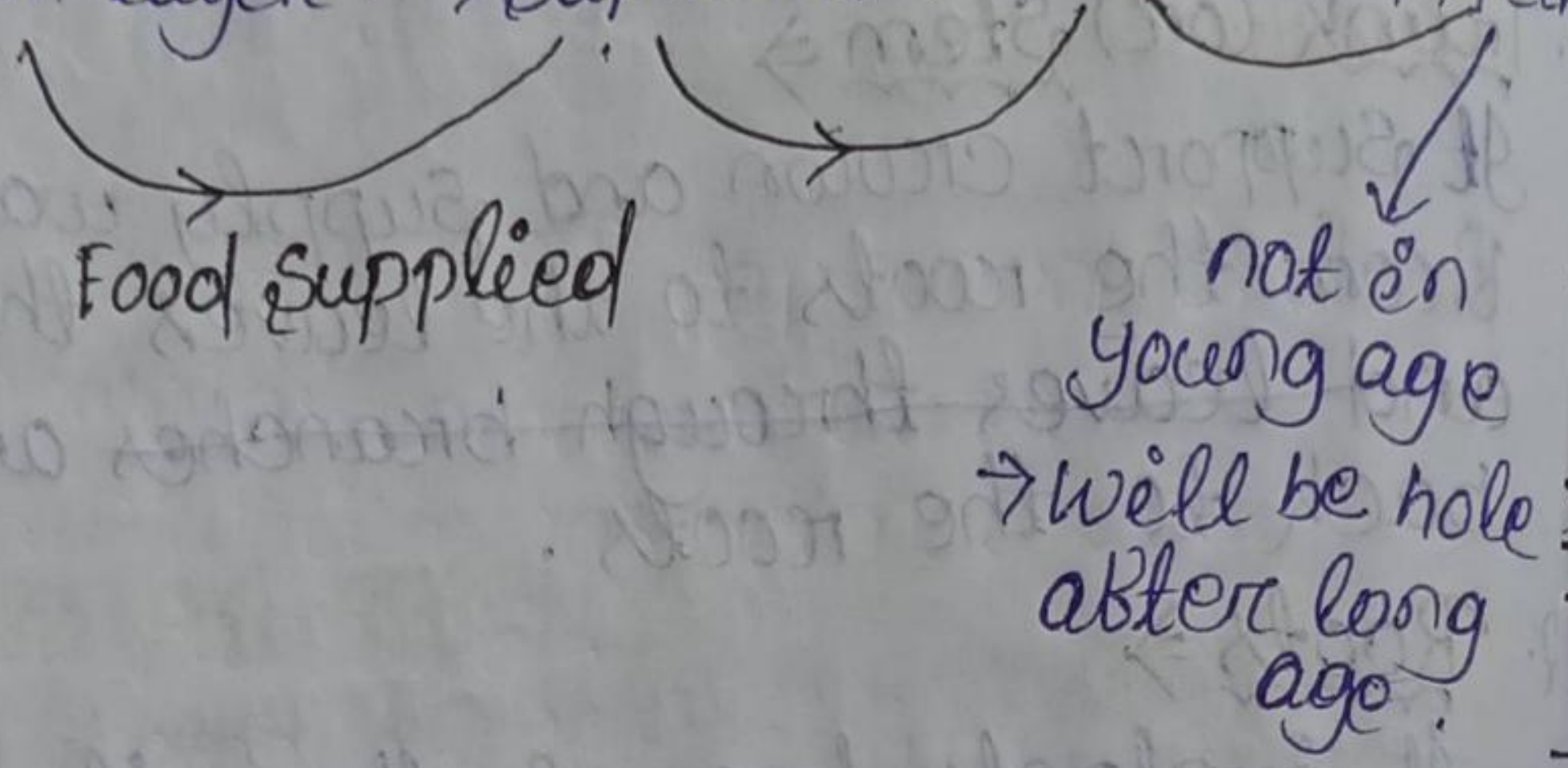
(a) crosssectional

(b) Longitudinal

(c) Radial.

Cross-sectional view of trunk of an exogenous trees:

Barks → cambium layer → Sapwood → Heartwood → Pith



Annual ring or growth ring:

- concentric layers of wood known as growth rings (or) annual rings.
- It consist of cellular tissue and woody fibre. These are formed due to the deposition of sap below bark. The no. of annual rings indicates the age of tree in a tropical climate.

1. Pith (or) Medulla.

- Innermost central portion (or) core of tree.
- It varies in size & shape for different types of trees.
- It consists entirely of cellular tissues & it nourishes the plant in its young age.
- * → It is larger in young trees than that in matured trees.
- Usually 12.5 mm dia. Sometimes barely visible.
- As the plant becomes old, the pith dies up and decays and the sap is then transmitted by the woody fibers deposited round the pith.
- The pith of branches is nothing but merely a prolongation of the pith of stem.

2. Heartwood:

- Inner annual rings surrounding the pith constitute the heartwood.
- Dark in colour generally & it is non porous & dense.

→ It indicates dead portion of tree and as such, it doesn't take active part in the growth of tree.

- * → It imparts rigidity to tree & hence it provides strong and durable timber for various engineering purpose.

3. Sap wood (or) Albunum:

- It outer annual rings betⁿ heart wood and cambium layer is known as sapwood.
- It is light in colour and weight. It shows recent growth and contains sap.

* → The annual rings of sap wood are less sharply

defined than those of heartwood. It takes active part in the growth of tree and the sap moves in an upward direction through it.

4. Cambium layer:

- The thin layer of sap betⁿ sap wood and inner bark is known as cambium layer.
- It indicates sap which has yet not been converted into sap wood.
- If the bark is removed for any reason the cambium layer gets exposed and the cells cease to be active resulting in death of tree.

[Bark is like our skin]

5. Innere bark:

- The inner skin or layer covering the cambium layer is known as the inner bark. It gives protection to the cambium layer from any injury.

Outer bark: [Cortex]

- The outer skin (or) cover of the tree is known as outer bark.
- It is outermost protective layer & sometime contains crack & fissure.
- It consists of cells of wood fibre & is also known as the cortex.

Medullary rays:

- The thin radial fibres extending from pith to cambium layer are known as the medullary rays.
- The function of these rays is to hold together the annual rings of heartwood and sapwood.

2 Microstructure

→ The structure of wood apparent only at great magnification is called the microstructure.

→ When studied under a microscope, it becomes evident that wood consists of living and dead cells of various shape & sizes.

* Living cell consists of 4 parts.

i. Membrane ⇒ consists of cellulose tissue and cellulose.

ii. Protoplasm ⇒ It is granular, transparent, viscous vegetable protein composed of carbon, hydrogen, oxygen, nitrogen & sulphur (CHONS)

(iii) Sap ⇒

(iv) Core ⇒ core of cell differs from protoplasm merely by the presence of phosphorus and it is generally oval.

The cells according to the functions they perform, are classified into 3 types.

1. Conductive cells: These cells serve mainly to transmit nutrients from roots to the branches and leaves.

2. Mechanical cells: These cells are elongated, thick-walled and have tightly interconnected narrow interior cavities.

These cells impart strength to the woods.

3. Storage cells: These cells serve to store & transmit nutrients to the living cells in the horizontal dirⁿ and they are usually located in medullary rays.

Seasoning of timber:

- Newly felled timber contains about 50% or more of its own dry weight as water. This water is in the form of sap and moisture. This water is to be removed before the timber can be used for any engineering purpose.
- In other words the process of drying of timber is called seasoning.
- If timber is used without seasoning, it is liable to shrink, warp and crack, it used may even rot and decay occurs.
- ∴ seasoning should be the first step for the efficient utilization of timber.

Objects of seasoning:

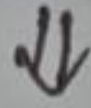
See properties

- * → To check/minimise the tendency of timber to shrink, warp & split.
- To increase the strength, durability and electrical resisting power of the timber (also to increase hardness, stickness).
- To increase the resisting power of timber as most of the causes of decay of timber are more or less related to timber seasoning.
- To maintain the shape and size of the timber articles which are expected to remain unchanged in form.
- To decrease the weight, so as to lower the cost of transport & handling.
- To make timber easily workable and to facilitate operations during conversion.
- To make timber suitable for gluing (i.e. effective joining of 2 members by glue).
- To make timber safe from attack of fungi and insects.

- To make timber fit for receiving treatment of paints, preservatives, varnishes etc.
- To allow timber to burn readily, it is used as fuel.

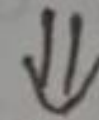
Method of Seasoning

(1) Natural Seasoning



Air seasoning

(2) Artificial Seasoning



(i) Boiling (ii) Chemical

(iii) electrical (iv) kiln

(v) water.

Properties of Good Timber:

Good quality timber has the following properties:

Colour: It should be uniform and dark.

Odour: It should be pleasant when freshly cut.

Sound: A clear ringing sound when struck, indicates the timber is good.

Texture: In good timber, the texture is fine and even.

Grains: Close grains indicate good timber.

Density: Higher the density, better is the timber.

Hardness: Harder timber is strong and durable.

Warping: Good timber retains its shape under changing environmental conditions.

Toughness: Timber should be capable of resisting shock loads.

Abrasion: Timber should be capable of resisting wear. This property is especially required if the timber is used for flooring.

Strength: Timber should have high strength in bending, shear and direct compression tension.

Modulus of elasticity: Timber with a high modulus of elasticity is preferred in constructing buildings.

Fire resistance: Good timber has high resistance to fire.

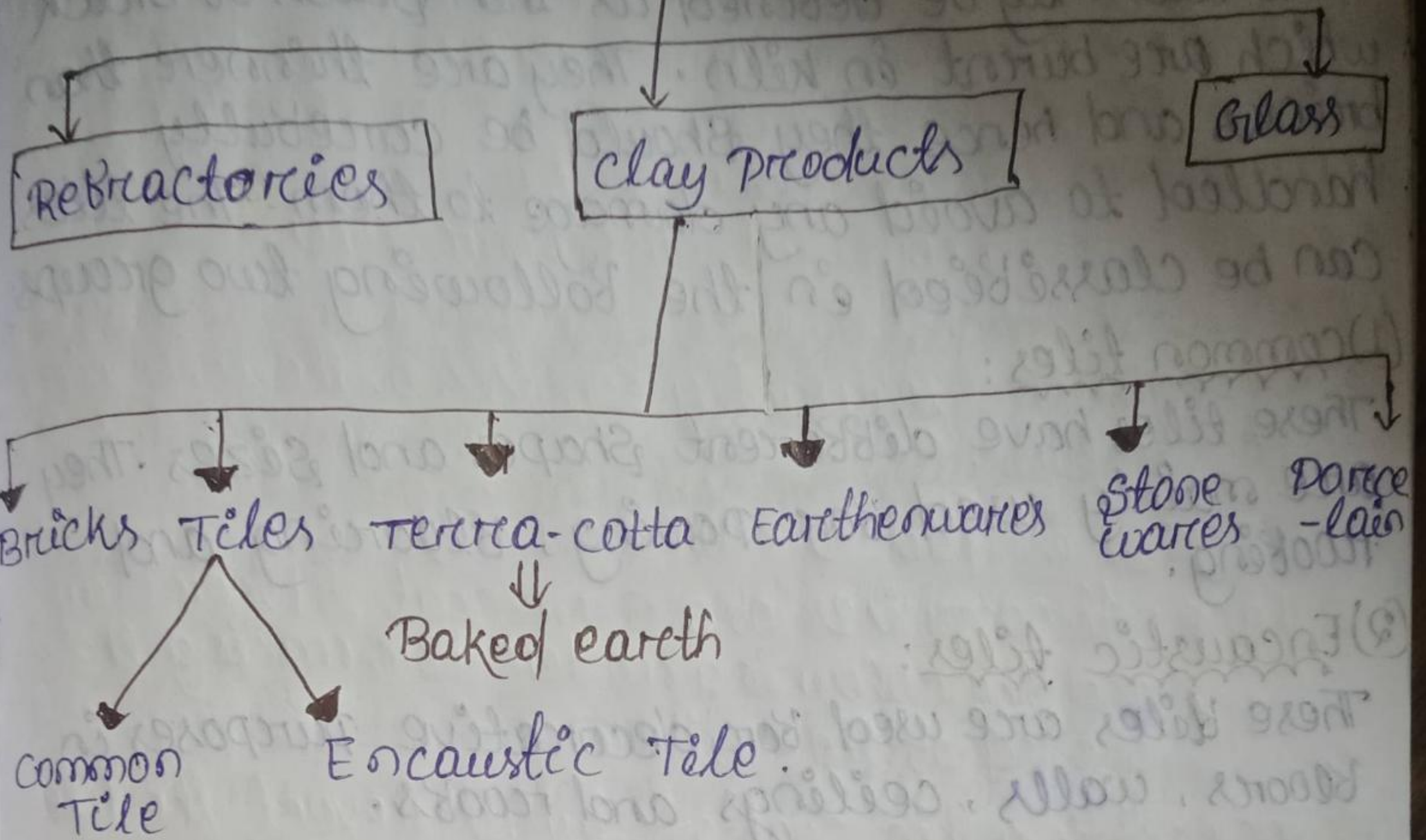
Permeability: Good timber has low water permeability. This is measured by the quantity of water filtered through a unit surface area of a specified thickness of wood.

Workability: Timber should be easily workable. It should not clog the saw and should be capable of being planed.

Durability: Good timber is capable of resisting the action of fungi, insects, chemicals and changing weather conditions. By seasoning and treating with preservatives, durability can be increased.

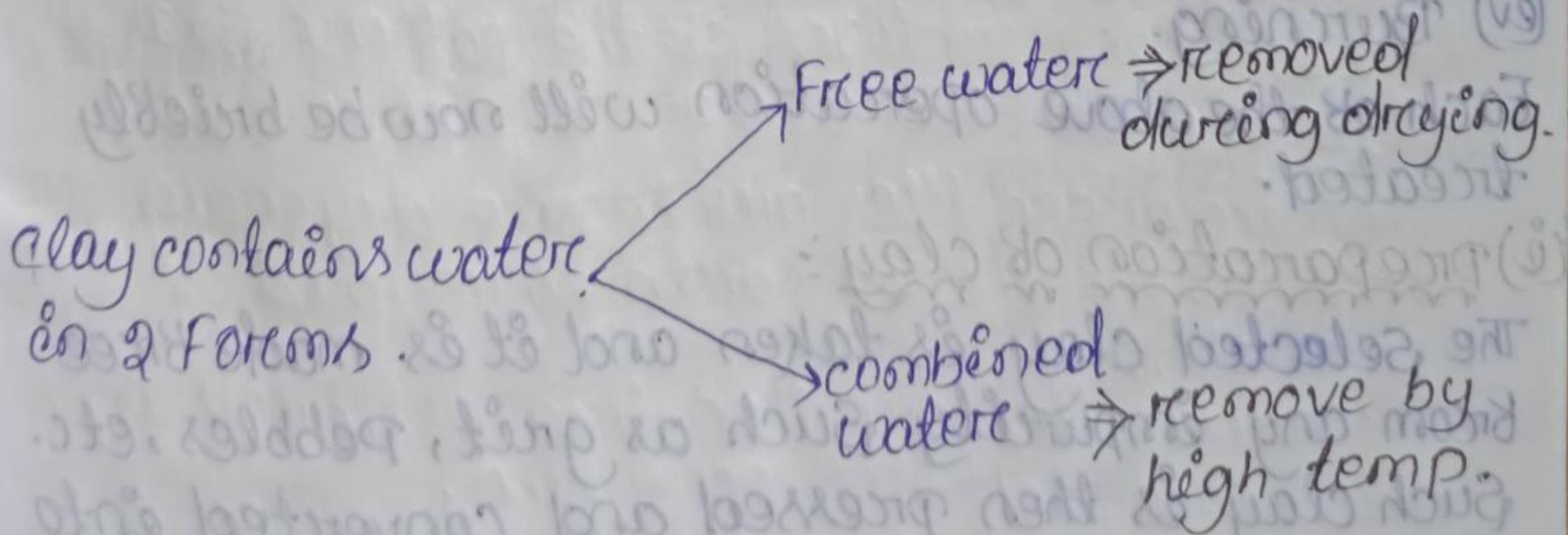
Defects: Good timber is free from serious defects like dead knots, shakes and cracks.

Ceramic Materials



Clay products :

Clay is a distinct product of chemical weathering of igneous rocks. Clay occurs plenty in nature. Orthoclase feldspar is mainly responsible for the production of clays in nature. This mineral, on decomposition, gives kaolinite which is free from iron oxide and alkalis.



Tiles:

The tiles may be defined as thin slabs of brick which are burnt in kiln. They are thinner than bricks and hence they should be carefully handled to avoid any damage to them. The tiles can be classified in the following two groups:

(1) Common tiles:

These tiles have different shapes and sizes. They are mainly used for paving, flooring and roofing.

(2) Encaustic tiles:

These tiles are used for decorative purposes in floors, walls, ceilings and roofs.

Common Tiles:

(1) Manufacture of common tiles:

Following four distinct operations are involved in the general process of manufacturing the common tiles.

(i) Preparation of clay

(ii) Moulding

(iii) Drying

(iv) Burning.

Each of the above operation will now be briefly treated.

(i) Preparation of clay:

The selected clay is taken and it is made free from any impurity such as grit, pebbles, etc. Such clay is then pressed and converted into fine powder in pug mills.

For tiles of superior quality, a large quantity of pure water is added to the powdered clay and it is well mixed in a tank. The mixture

is then allowed to stand quietly. The coarse heavy particles settle at the bottom of tank. The fine is then allowed to dry off. The fine clay left after such process is used for the manufacture of tiles.

To make the tiles hard and impervious, a mixture of ground glass and potteryware may be added in required quantity to the clay of tiles.

(ii) Moulding:

The clay is placed in moulds which represent the pattern or shape in which the tile is to be formed. The moulding may be done either with the help of wooden moulds or mechanical means or potter's wheel.

The wooden moulds should be prepared from well seasoned timber. The clay is pressed into such moulds and tiles are ready for drying when clay is taken out of moulds. The care should be taken to preserve the shape of tiles during the removal of moulds. The tiles which do not have a uniform section throughout their length are moulded with the help of wooden moulds.

The moulding with the help of mechanical means includes the provision of machines and the clay is pressed into such machines to get tiles of

Encaustic Tiles:

The encaustic tiles are manufactured from carefully prepared ordinary clays, colouring materials and sometimes with finer clays. Depending upon the colouring pigment added in the clay, these tiles obtain the desired print or colour after manufacture.

An encaustic tile usually consists of the following three layers.

(1) Body: It is made a coarser clay.

(2) Face: It comprises of a 6 mm coat of finer clay and the colouring matter for making the ground of the pattern.

(3) Back: It is a thin coat of clay to prevent the tile from warping.

The manufacturing process of these tiles is as follows:

(i) The face is moulded to the desired pattern.

(ii) The coarser clay body is put on the back of the face and also a thin coat to form the back.

(iii) The maker's name is stamped on the back.

(iv) A few holes are kept for joining with cement during laying.

(v) The clay with different colours is poured into the different portions of the pattern so as to obtain the desired design of colours.

(vi) When the green tiles are become dry enough for handling, the excess earth is removed or scraped off.

(vii) The green tiles are then trimmed, dressed, dried and burnt in the dome kiln.

(viii) If required, the burnt tiles may be glazed by dipping them into a mixture of powdered glass and water and then reheating them.

Terra-cotta :-

(1) General : The terra means earth and cotta means baked. Hence the terra-cotta means the baked earth. It is thus a type of earthenware or porous pottery made from local clays and glazed with glazes containing galena. It is soft enough to be scratched by a knife.

(2) Manufacture of terra-cotta :

Following four distinct operations are involved in the manufacture of terra-cotta :

(i) Preparation of clay

(ii) Moulding

(iii) Drying

(iv) Burning

Each of the above operation will now be briefly described.

(i) Preparation of clay : For terra-cotta, the selected clay is taken. The clay should contain a slightly higher percentage of iron oxide, about 5% to 8% and proportion of lime should be less, about 1 percent or so. Sometimes several varieties of clay with high alumina content are taken and then to this mixture is added sand, ground glass, old terra-cotta or pottery. The addition of

Such materials gives strength and rigidity to the terra-cotta products and it prevents shrinking while drying.

Such clay is made free from any impurity such as grit, pebbles, organic matter, etc. It is then finely crushed and pulverized. The water is added in required quantity and the ingredients are thoroughly mixed with spades. Such wet clay is kept for several days in a damp condition for weathering and tempering. It is then pressed

or kneaded in a pug mill and it is made ready for the next operation of moulding. The required quantity of colouring substance is added at this stage to obtain the desired shade of colour in the final product of terra-cotta.

(ii) Moulding:

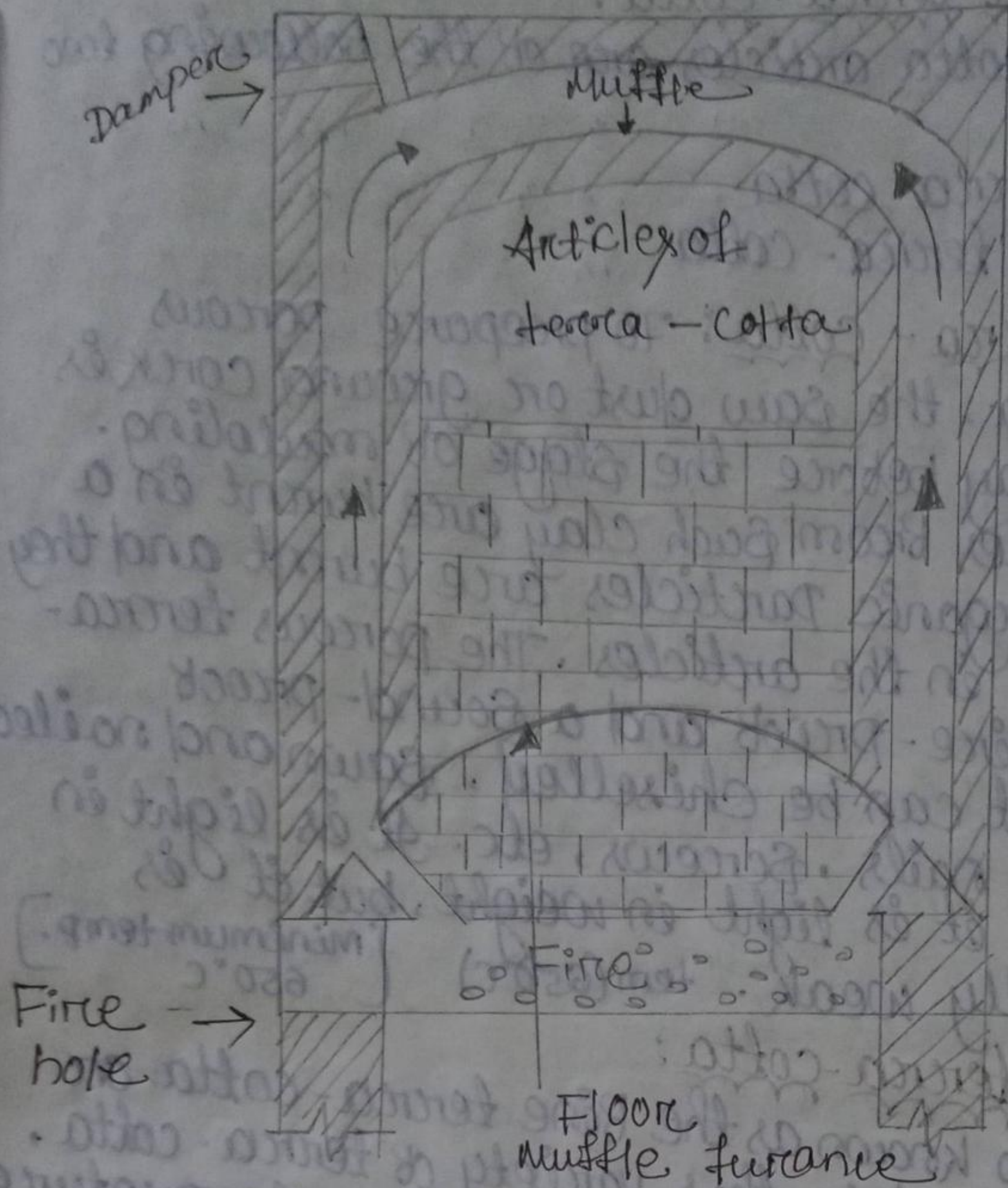
The clay is placed in moulds which represent the pattern or shape in which the product is to be formed. For terra-cotta work, special moulds of plaster or paries or templates of zinc are used. The size of moulds is determined by keeping due allowance for shrinkage. The fine sand is sprinkled on the inside surface of moulds and clay is then pressed in moulds with hand.

(iii) Drying:

The moulds filled in with clay are kept for some days for drying. After this period, the articles of the terra-cotta are taken out from the mould and they are allowed to dry further in a room or under a shed. The drying should be done carefully and slowly with proper control of temperature. The gradual drying helps in retaining the correct shape and size of the blocks.

(iv) Burning:

The dried products are then burnt in special muffle furnaces. temp (650-1200°C), 4 days burning & 5 days cooling down.



A muffle indicates a box or a compartment of a furnace in which things can be heated without contact with the fuel and its products. A dampener indicates a metal plate which is provided in an opening to regulate the draught. The dried articles are arranged in muffle and temperature of kiln is raised to about 1200°C . This temperature is maintained for about four days and the burnt products are then allowed to cool down in kiln for a period of about five days. For getting the glazed products, the glazed materials should be applied by brush on terra-cotta products before they are burnt.

(3) varieties of terracotta:

The terracotta articles are of the following two types:

- (i) porous terracotta
- (ii) polished terracotta

(i) porous terracotta: To prepare porous terracotta, the saw dust or ground cork is added in clay before the stage of moulding. When articles from such clay are burnt in a kiln, the organic particles are burnt and they leave pores in the articles. The porous terracotta is a fire-proof and a sound-proof material. It can be chiselled, sawn and nailed easily with nails, screws, etc. It is light in weight, but it is light in weight, but it is structurally weak. $tem(650^{\circ}C)$ [minimum temp.] $650^{\circ}C$

(ii) polished terracotta:

This is also known as the bone terracotta or Baience. To obtain this variety of terracotta, the articles are burnt at a lower temperature of about $650^{\circ}C$. The first burning is known as the biscuiting. The articles brought to biscuit stage are removed from kiln and are allowed to cool down. They are then coated with glazing compound and burnt again in the kiln at a temperature of about $1200^{\circ}C$. The finish is available in a variety of colours and it indicates superior quantity of terracotta. It is used for ornamental purpose and in industrial areas since it is ordinarily unaffected by the adverse atmospheric conditions.

(4) Advantages of terracotta:

Following are the advantages of terracotta:

- (i) It is strong and durable material.
- (ii) It is available in different colours.
- (iii) It is cheaper than ordinary finely dressed stones.
- (iv) It is easily cleaned.
- (v) It is easily moulded in desired shapes.
- (vi) It is fire-proof and can therefore be conveniently used with R.C.C. work.
- (vii) It is light in weight.
- (viii) It is not affected by atmospheric agencies and acids and is capable of withstanding weathering actions better than most kinds of stone.

(5) Disadvantages of terracotta:

- (i) It cannot be fixed during the progress of work. But it is to be fixed when the work is in final stage of completion.
- (ii) It is twisted due to unequal shrinkage in drying and burning.

(6) Uses of terracotta:

- (i) The hollow terracotta blocks are used for various ornamental purposes such as facing work, arches, cornices, casing for columns, etc.
- (ii) It is adopted for all sorts of ornamental work.
- (iii) It is used as a decorative material in place of stones for ornamental parts of buildings such as cornices, string courses, sills, copings, bases of pillars, fire places, etc.

Earthenware:

The term earthenware is used to indicate wares or articles prepared from clay which is burnt at low temperature and cooled down slowly. The clay

is mixed with required quantity of sand, crushed pottery, etc. The addition of such materials prevents the ~~during~~ shrinkage during drying and burning.

The earthenwares are generally soft and porous. When glazed, the earthenwares become impervious to the water and they are not affected by acids or atmospheric agencies. The terra-cotta is a kind of earthenware. The earthenware is used for making ordinary drain pipes, electrical cable conduits, partition blocks, etc.

Stoneware:

The term stoneware is used to indicate the wares or articles prepared from refractory clays which are mixed with stone and crushed pottery. Such a mixture is then burnt at a high temperature and cooled down slowly.

The stoneware is more compact and dense than earthenware. When glazed, the stonewares become impervious to the water and they are not affected by acids or atmospheric agencies.

The sound stonewares give clear ringing sound when struck with each other.

The stonewares are strong, impervious, durable and resistant to corrosive fluids and they resemble fire bricks. The stonewares can be kept clean easily and hence they have become very popular as the sanitary articles such as wash basins, sewer pipes, glazed tiles, water closets, gully traps etc. They are also used as jars to store the chemicals.

Porcelain:

The term porcelain is used to indicate fine earthenware which is white, thin and semi-transparent. Since the colour of porcelain is white, it is also referred to as the whiteware.

High voltage Porcelain

Name	Properties	Uses
carbon and graphite	It is a refractory material of high quantity. But it is oxidized at high temperature.	It is used for making electrodes & in the construction of atomic reactor rockets.
carbon brick	It is prepared from powder coke and tar. It can resist high temperature.	It is used as lining material refractory bricks, etc. for electric furnace.
cordierite porcelain	It contains 22% alumina, 35% clay and 43% silicate of magnesia. It is available in porous, partly porous and glassy form.	It is used for electric furnace, refractory bricks, etc.
steatic porcelain	It contains 70 to 90% silicate of magnesia.	It is used as electrical insulator for high intensity electric current, vacuum tubes, etc.
zircon porcelain	It contains 45 to 60% zircon, 15 to 30% clay and 15 to 30% silicate of zircon. Its dielectric constant at high temperature is good.	It is used in the manufacture of spark plugs.

The clay of sufficient purity and possessing high degree of tenacity and plasticity is used in preparing porcelains. It is hard, brittle and non-porous. It is prepared from clay, kalspar, quartz and minerals. The constituents are finely ground and then they are thoroughly mixed in liquid state. The mixture is given the desired shape and it is burnt at high temperature.

(1) General :

The surface of clay products are sometimes glazed. A glaze is a glassy coat of thickness about 0.1 to 0.2 mm applied on the surface of an item and then fused into place by burning at high temperature.

(2) Purpose :

Following are the purposes for which the glazing is done :

- (i) To improve the appearance.
- (ii) To make the articles durable and impervious.
- (iii) To produce the decorative effects.
- (iv) To protect the articles from action of atmospheric agencies, chemicals, sewage, etc.
- (v) To provide smooth surface.

(3) Methods of glazing:

The glazing may be transparent like glass or it may be opaque like enamels. For obtaining coloured glazes, the oxides and salts of various metals or special refractory colouring agents are added. For instance, the addition of copper oxides will impart green colour and addition of iron oxide will impart red and brown colours.

(i) Transparent glazing:

This type of glazing may be given by the following two methods:

- (a) Salt glazing
- (b) Lead glazing

(a) Salt glazing: In this method, a small quantity of wet sodium chloride or salt is added in the kiln at a high temperature of about 1300°C . The salt is

vaporized at a high temperature and a glass like glaze is formed on the surface of articles due to sticking of vapour of salt. This method is useful for sanitary pipes and chemical stonewares. The quantity of wet salt and throwing it at proper time should be done with extreme care. The colour of articles glazed by this method is brownish.

(b) Lead glazing:

For getting articles of better quality, the lead glazing is preferred to the salt glazing. In this method, the article is once burnt and it is then dipped in a bath containing oxide of lead and tin. The article is taken out from the bath and it is reburnt at a high temperature. The particles of oxide of lead and tin melt and they form a film of glass over the exposed surfaces of the article. In this method of glazing, the glaze does not penetrate into the body of ware and as a matter of fact, it can easily be detached from the ware surface. This method is used for terra-cotta, bone-clay wares and earthenwares.

(ii) opaque glazing:

This type of glazing is adopted to give better appearance than that given by the burnt material. The superior clay is finely powdered and dried. The sufficient quantity of water is added to such clay to make a plastic cream like substance, known as the slip.

The articles to be glazed are dipped in the slip before burning and they are subsequently heated. The burning of articles results into the flow of clay particle and an opaque glaze surface is formed. The sanitary articles are glazed by this

Process:

Properties of cast-iron:

- Following are the properties of cast-iron:
- (i) It is placed in salt water, it becomes soft.
 - (ii) It can be hardened by heating and sudden cooling, but it cannot be tempered.
 - (iii) It cannot be magnetised.
 - (iv) It does not rust easily.
 - (v) It is fusible.
 - (vi) It is hard, but it is brittle also.
 - (vii) It is not ductile and hence it cannot be adopted to absorb shocks and impacts.
 - (viii) Its melting temperature is about 1250°C .
 - (ix) It shrinks on cooling. This fact is to be considered while making patterns or moulds.

For boundary work:

(x) Its structure is granular and crystalline with whitish or greyish tinge.

(xi) Its specific gravity is 7.5.

(xii) It lacks plasticity and hence it is unsuitable for the forging work.

(xiii) It is weak in tension and strong in compression. The tensile and compressive strengths of cast-iron of average quality are respectively 150 N/mm^2 (tension) and 600 N/mm^2 (compression).

(xiv) The two pieces of cast-iron cannot be connected by the process of riveting or welding. They are to be connected by nuts and bolts which are fixed to the flanges. The holes for bolts, etc. are either drilled out or cast in the casting.

Uses of cast-iron:

The use of cast-iron is not recommended in horizontal direction either for heavy or variable loads or at places where there are chances for the slightest shock to exist. The cast-iron cracks and snaps suddenly when subjected to the

works, overloading or vice without giving any warning of approaching failure under such stresses. The cast iron to be used on the works should be tough, close grained grey metal. Free from air holes, sand holes, blows and with an even surface. It should be sufficiently soft to admit of being easily cut either by a chisel or a drill.

Following are the important uses of cast-iron:

- (i) For making cisterns, water pipes, gas pipes and sewers, manhole covers and sanitary fittings.
- (ii) For making ornamental casting such as brackets, gates, lamp posts, spiral staircase, etc.
- (iii) For making parts of machinery which are not subjected to heavy loads.
- (iv) For manufacturing compression members like columns in building, bases of columns, etc.
- (v) For preparing agricultural implements.
- (vi) For preparing rail chairs, carriage wheels, etc.

Properties of wrought-iron:

Following are the properties of wrought-iron.

- (i) It becomes soft at white heat and it can be easily forged and welded.
- (ii) It can be used to form temporary magnets, but cannot be magnetised permanently.
- (iii) It fuses with difficulty. It cannot therefore be adopted for making castings.
- (iv) It is ductile, malleable and tough.
- (v) It is moderately elastic.
- (vi) It is unaltered by saline water.
- (vii) It resists corrosion in a better way.
- (viii) Its fresh fracture shows clear bluish colour with a high silky luster and fibrous appearance.

- (ix) Its melting point is about 1500°C .
- (x) Its specific gravity is about 7.8.
- (xi) Its ultimate compressive strength is about 200N/mm^2 .
- (xii) Its ultimate tensile strength is about 400N/mm^2 .

Defects in wrought-iron:

The wrought-iron which has become defective may either be cold short or red short. The cold short wrought-iron is very brittle when it is cold - it cracks, is bent. It may however be worked at high temperature. This defect occurs when phosphorus is present in excess quantity.

The red short wrought-iron possesses sufficient tenacity when cold. But it cracks when bent or finished at a red heat. It is therefore useless for welding purpose. This defect occurs when sulphur is present in excess quantity.

Use of wrought-iron:

The wrought iron is replaced at present to a very great extent by mild steel. It is therefore produced to a very small extent at present. It is used where a tough material is required.

The wrought-iron, at present, is used for rivets, chains, ornamental iron work, railway couplings, water and steam pipes, raw material for manufacturing, steel, bolts and nuts, horse shoe bars, handrails, straps for timber roof trusses, boiler tubes, roofing sheets, armatures, electro-magnets etc.

Uses of steel

Name of steel	Carbon content	Uses
Mild steel	up to 0.10%	Motor body, sheet metal, tinplate etc.
Medium carbon steel	up to 0.25%	Boiler plates, structural steel etc.
High carbon steel or hard steel	up to 0.45%	Rails, tyres, etc.
	up to 0.60%	Hammers, large stamping and pressing dies, etc.
	up to 0.75%	Sledge hammers, springs, stamping dies, etc.
	up to 0.90%	Miner's drills, Smith's tools, stone mason's tools, etc.
	up to 1.00%	chisels, hammers, saws, wood working tools, etc.
	up to 1.10%	Axes, cutlery, drills, knives, picks, punches etc.

Properties of Mild-steel:

Following are the properties of mild steel:

- (i) It can be magnetised permanently.
- (ii) It can be readily forged and welded.
- (iii) It cannot be easily hardened and tempered.
- (iv) It has fibrous structure.
- (v) It is malleable and ductile.
- (vi) It is not easily attacked by salt water.
- (vii) It is tougher and more elastic than wrought iron.

- (viii) It is used for all types of structural work.
- (ix) It rusts easily and rapidly.
- (x) Its melting point is about 1400°C .
- (xi) Its specific gravity is 7.80.
- (xii) Its ultimate compressive strength is about 80 to 120 kN per cm^2 or $(800 - 1200 \text{ N/mm}^2)$.
- (xiii) Its ultimate tensile and shear strengths are

about 60 to 80 kN per cm^2 .

Properties of Hard Steel:

- (i) It can be easily hardened and tempered.
- (ii) It can be magnetised permanently.
- (iii) It cannot be readily forged and welded.
- (iv) It has granular structure.
- (v) It is not easily attacked by salt water.
- (vi) It is tougher and more elastic than mild steel.
- (vii) It is used for finest cutlery, edge tools and core parts which are to be subjected to shocks and vibrations.
- (viii) It rusts easily and rapidly.
- (ix) Its melting point is about 1300°C .
- (x) Its specific gravity is 7.90.
- (xi) Its ultimate compressive strength is about 140 to 200 kN per cm^2 .
- (xii) Its ultimate shear strength is about 110 kN per cm^2 .
- (xiii) Its ultimate tensile strength is about 80 to 110 kN per cm^2 .