

LESSON PLAN OF 5th SEMESTER(2022-23) CHEMICAL ENGINEERING		
DISCIPLINE: CHEMICAL	Semester:-5TH	NAME OF THE TEACHING FACULTY PRATEEK KUMAR DAS
SUBJECT: CHEMICAL ENGINEERING THERMODYNAMICS	No of days per Week Allotted : 04	SEMESTER: SEPTEMBER TO DECEMBER No of Weeks:- 15
Week	Class/ Day	Theory/ Practical Topics
1 ST	1 st	Scope and limitations of Thermodynamics
	2 nd	System, surrounding and boundary
	3 rd	Different types of systems
	4 th	Processes, state, properties
2 ND	1 st	Path and State functions
	2 nd	Heat and Work
	3 rd	Equilibrium state and phases
	4 th	Zeroth law of Thermodynamics
3 rd	1 st	State and explain first law of Thermodynamics
	2 nd	State and explain first law of Thermodynamics
	3 rd	Concept of internal energy, Enthalpy, heat capacity
	4 th	Concept of internal energy, Enthalpy, heat capacity
4 th	1 st	First law of thermodynamics for cyclic process, non-flow process, and flow process
	2 nd	First law of thermodynamics for cyclic process, non-flow process, and flow process
	3 rd	First law of thermodynamics for cyclic process, non-flow process, and flow process
	4 th	Solve numerical on application of 1ST law of thermodynamics
5 th	1 st	Solve numerical on application of 1ST law of thermodynamics
	2 nd	Constant volume process for ideal gases
	3 rd	Constant pressure process for ideal gases
	4 th	Constant temperature process for ideal gases
6 th	1 st	Adiabatic process for ideal gases
	2 nd	Polytropic process for ideal gases
	3 rd	Solve simple problems
	4 th	Solve simple problems
7 th	1 st	Solve simple problems
	2 nd	Equation of state and ideal gas
	3 rd	P-V-T behavior of pure fluid
	4 th	P-V-T behavior of pure fluid

8 th	1 st	Concept of heat reservoir, heat engine, and heat pump
	2 nd	Concept of heat reservoir, heat engine, and heat pump
	3 rd	State and explain second law of thermodynamics
	4 th	Concept of entropy
9 th	1 st	Concept of entropy
	2 nd	Calculate change of entropy for various conditions
	3 rd	Calculate change of entropy for various conditions
	4 th	Calculate change of entropy for various conditions
10 th	1 st	Third law of Thermodynamics
	2 nd	Solve simple problems
	3 rd	Solve simple problems
	4 th	Classify thermodynamic properties
11 th	1 st	Work function and Gibb's free energy
	2 nd	Work function and Gibb's free energy
	3 rd	Gibb's phase rule
	4 th	Various relationships among thermodynamic properties
12 th	1 st	Maxwell equation
	2 nd	Maxwell equation
	3 rd	Clapeyron equation
	4 th	Entropy-heat capacity relation
13 th	1 st	Differential equation for entropy
	2 nd	Effect of temperature, pressure and volume on U,H and S, relationship between Cp and Cv
	3 rd	Effect of temperature, pressure and volume on U,H and S, relationship between Cp and Cv
	4 th	Gibb's-Helmholtz equation
14 TH	1 st	Fugacity co-efficient, effect of temperature and pressure on fugacity, fugacity of pure gases, solids and liquids
	2 nd	Fugacity co-efficient, effect of temperature and pressure on fugacity, fugacity of pure gases, solids and liquids
	3 rd	Concept of activity, Effect of pressure and temperature on activity
	4 th	Concept of activity, Effect of pressure and temperature on activity
15 TH	1 st	Concept of Refrigeration and liquefaction process
	2 nd	Objective Questions discussion
	3 rd	Objective Questions discussion
	4 th	Objective Questions discussion